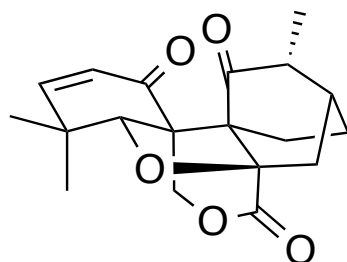


Enantioselective Total Syntheses of (-)-Maoecrystal V

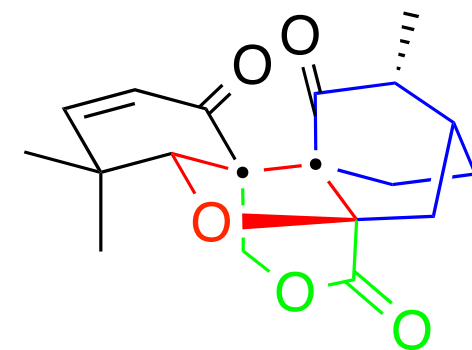
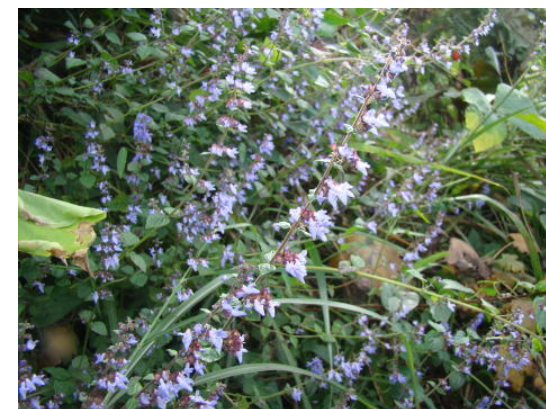
J. Am. Chem. Soc. **2014**, *136*, 17738–17749.
(DOI: 10.1021/ja510573v)

And

J. Am. Chem. Soc. **2014**, *136*, 17750–17756.
(DOI: 10.1021/ja5109694)



About Maoecrystal V

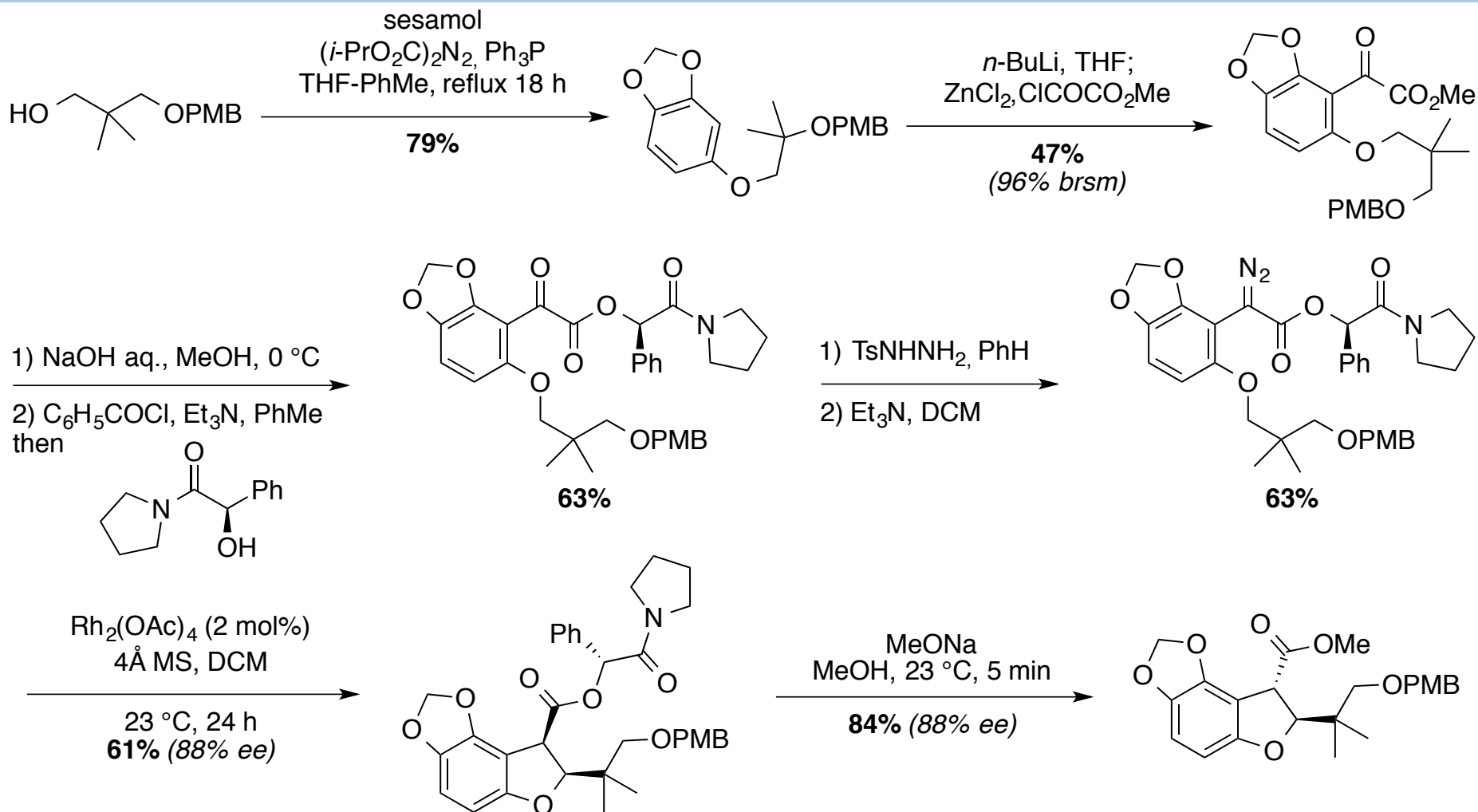


- > Isolated from *Isodon eriocalyx* (Sun & co, 1994)
- > Structure assignment *via* crystal X-ray diffraction (2004)
- > 2 vicinal quaternary center, Strained cyclic ether, a spirolactone, a bicyclo[2.2.2]octane ring system and 6 stereogenic centers
- > Cytotoxic activity against HeLa cells (IC₅₀ = 60 nM)

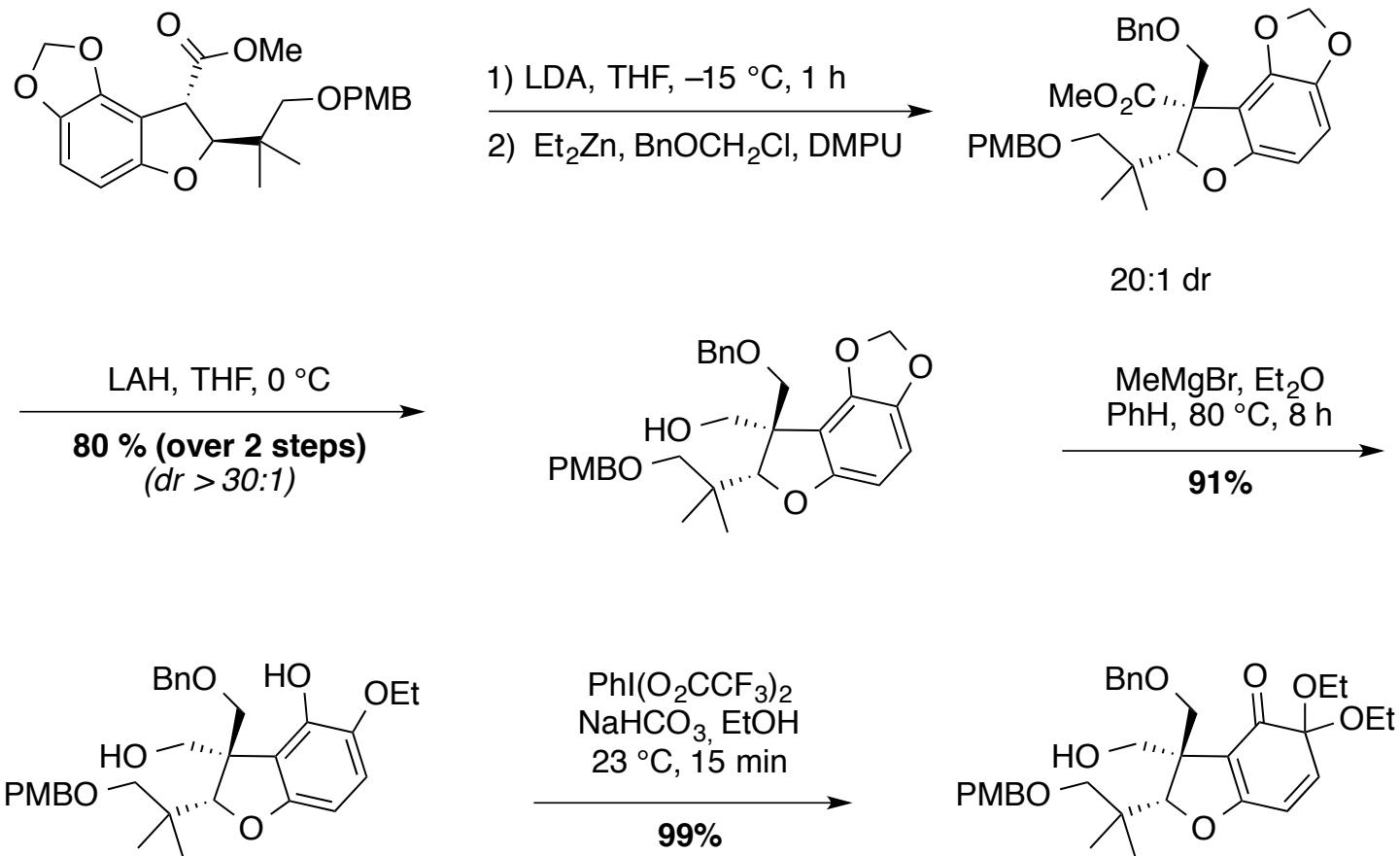
Sheng-Hong Li; Jia Wang; Xue-Mei Niu; Yun-Heng Shen; Hong-Jie Zhang; Han-Dong Sun; Ma-Lin Li; Qin-E Tian; Yang, L.; Peng Cao, A.; Zheng, Q.-T. *Org. Lett.* **2004**, *6*, 4327–4330.

Sun, H.-D.; Huang, S.-X.; Han, Q.-B. *Nat. Prod. Rep.* **2006**, *23*, 673–698.

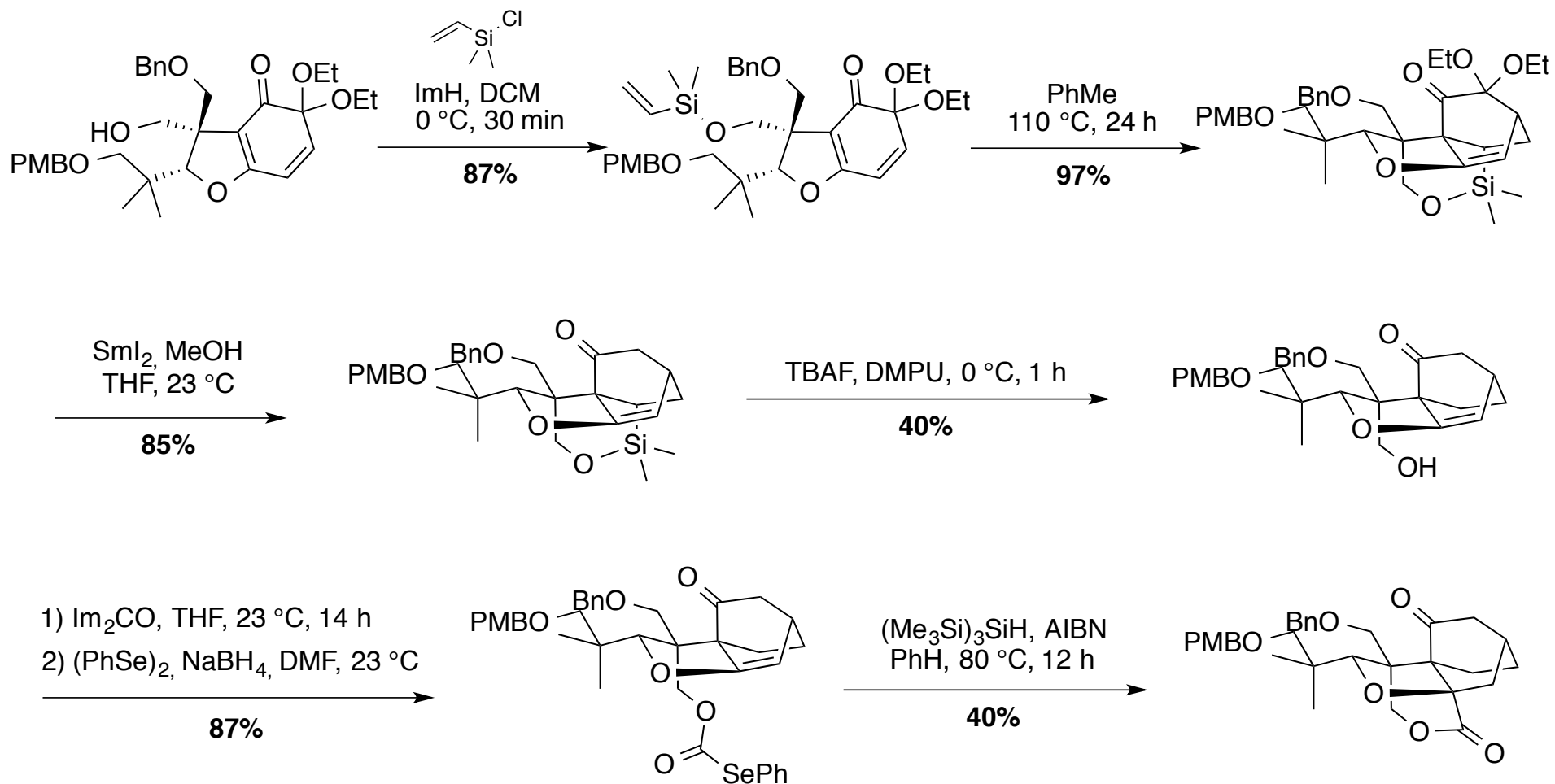
Zakarian/Davies' Synthesis



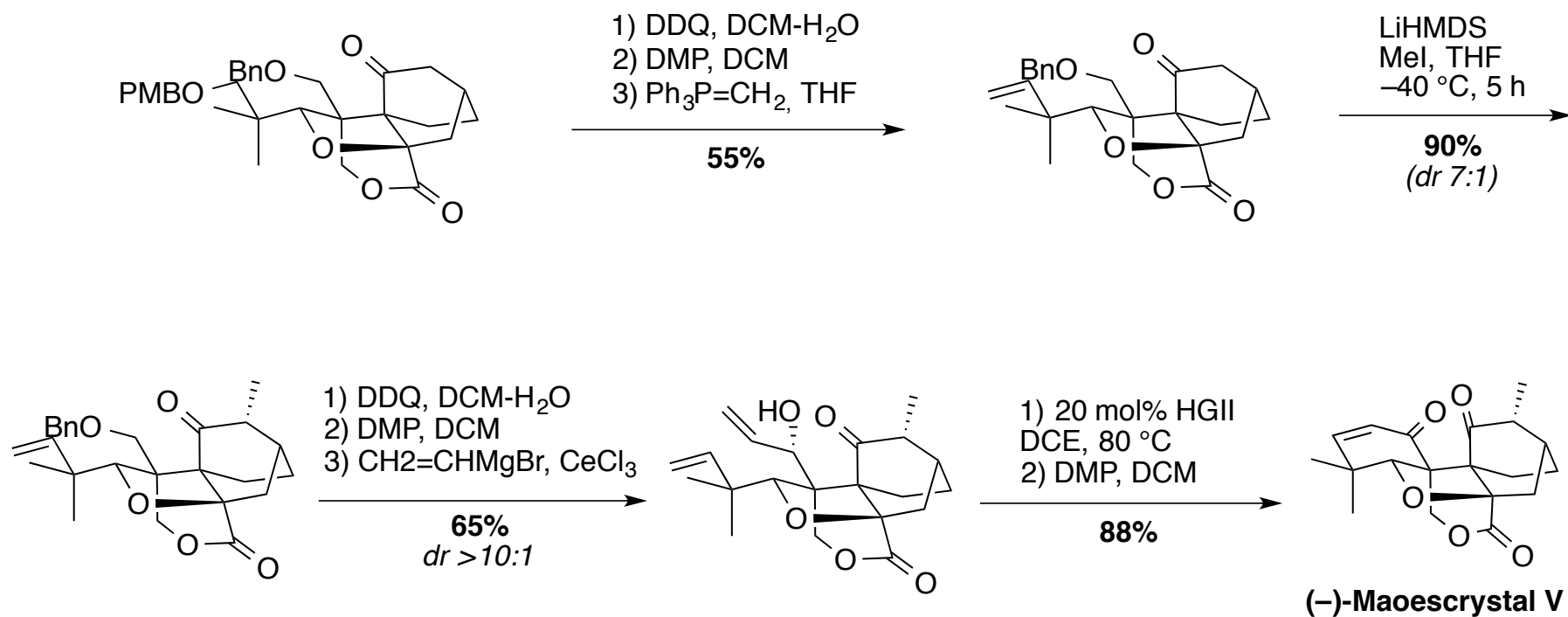
Zakarian/Davies' Synthesis



Zakarian/Davies' Synthesis



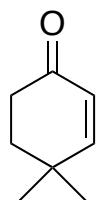
Zakarian's Synthesis



Thomson's Synthesis

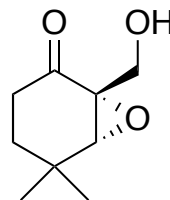
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UNIVERSITÄT
BERN



1) CH₂O, sodium dodecyl sulfate
DMAP, H₂O

2) (-)-DIPT, Ti(O*i*Pr)₄
*t*BuOOH, 4Å MS



(53% over 2 steps)
**Sharpless
epoxidation
94% ee**

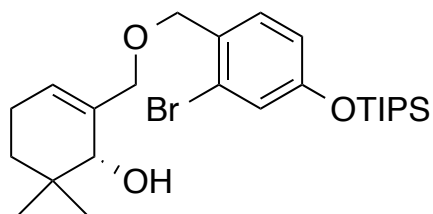
3) TfOH, **10**

4) NaBH₄

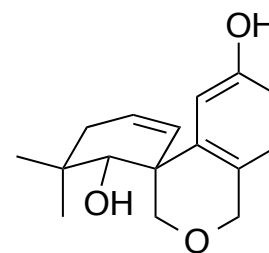
5) I₂, Ph₃P, imidazole

6) Zn, NH₄Cl

(61% over 4 steps)

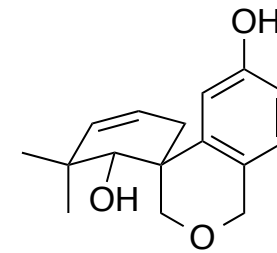


7) TESCl, imidazole (92%)
8) Pd(PPh₃)₄, PMP, 125 °C then
TBAF

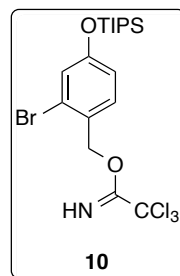


**minor
13%**

+

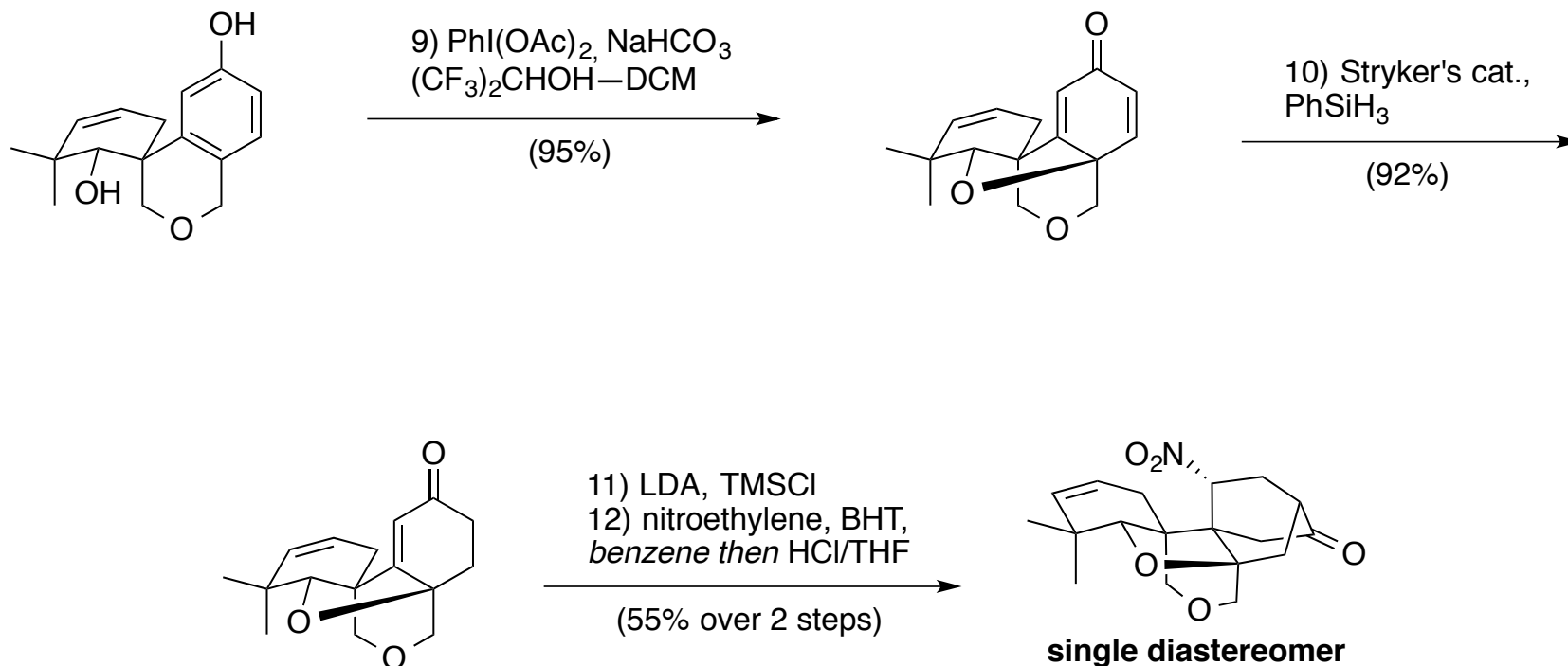


**major
52%**

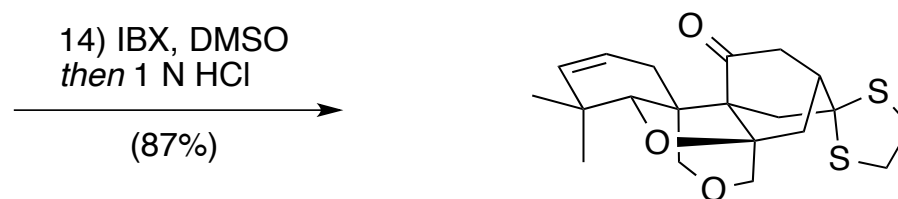
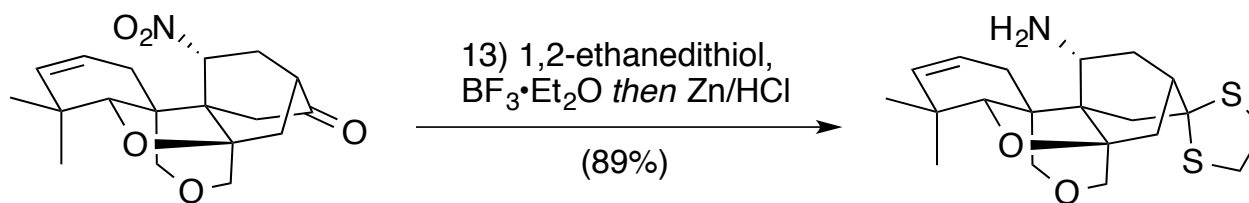


Thomson's Synthesis

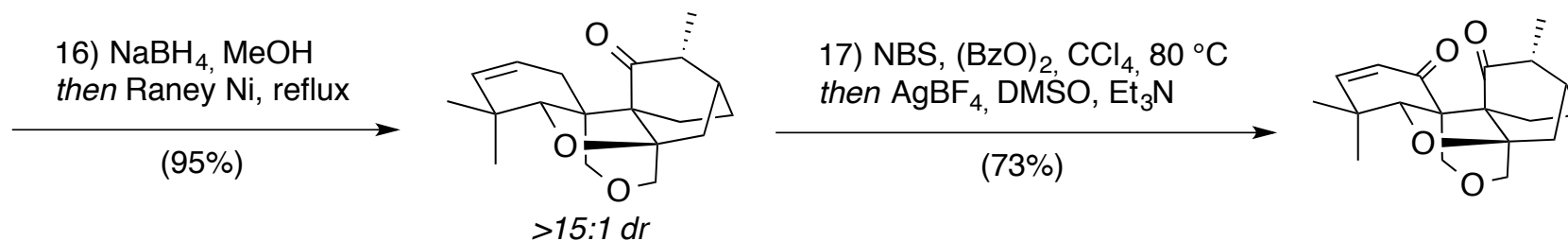
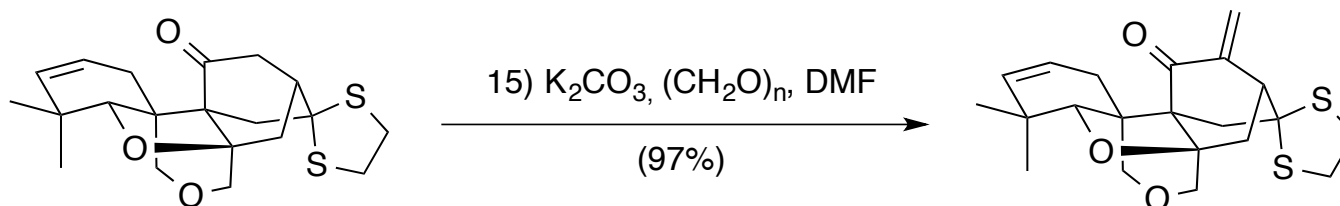
Stryker's cat. = $[(\text{Ph}_3\text{P})\text{CuH}]_6$



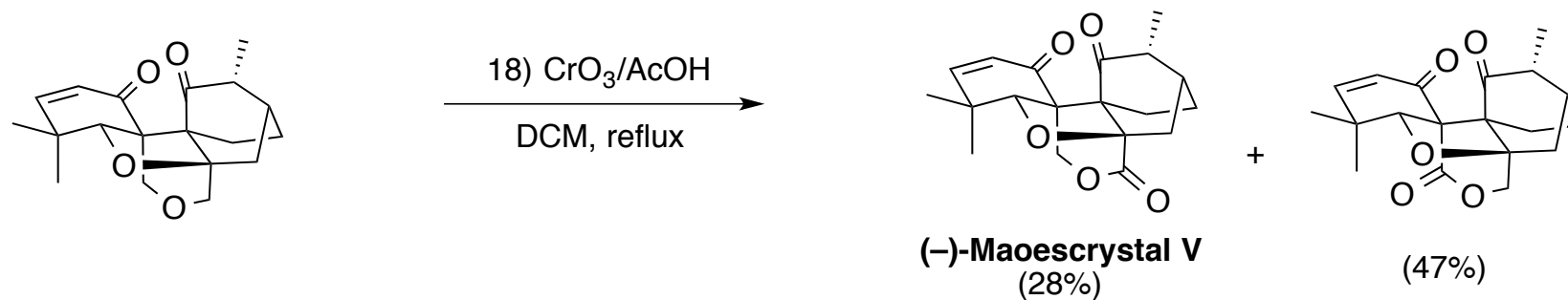
Thomson's Synthesis



Thomson's Synthesis



Thomson's Synthesis



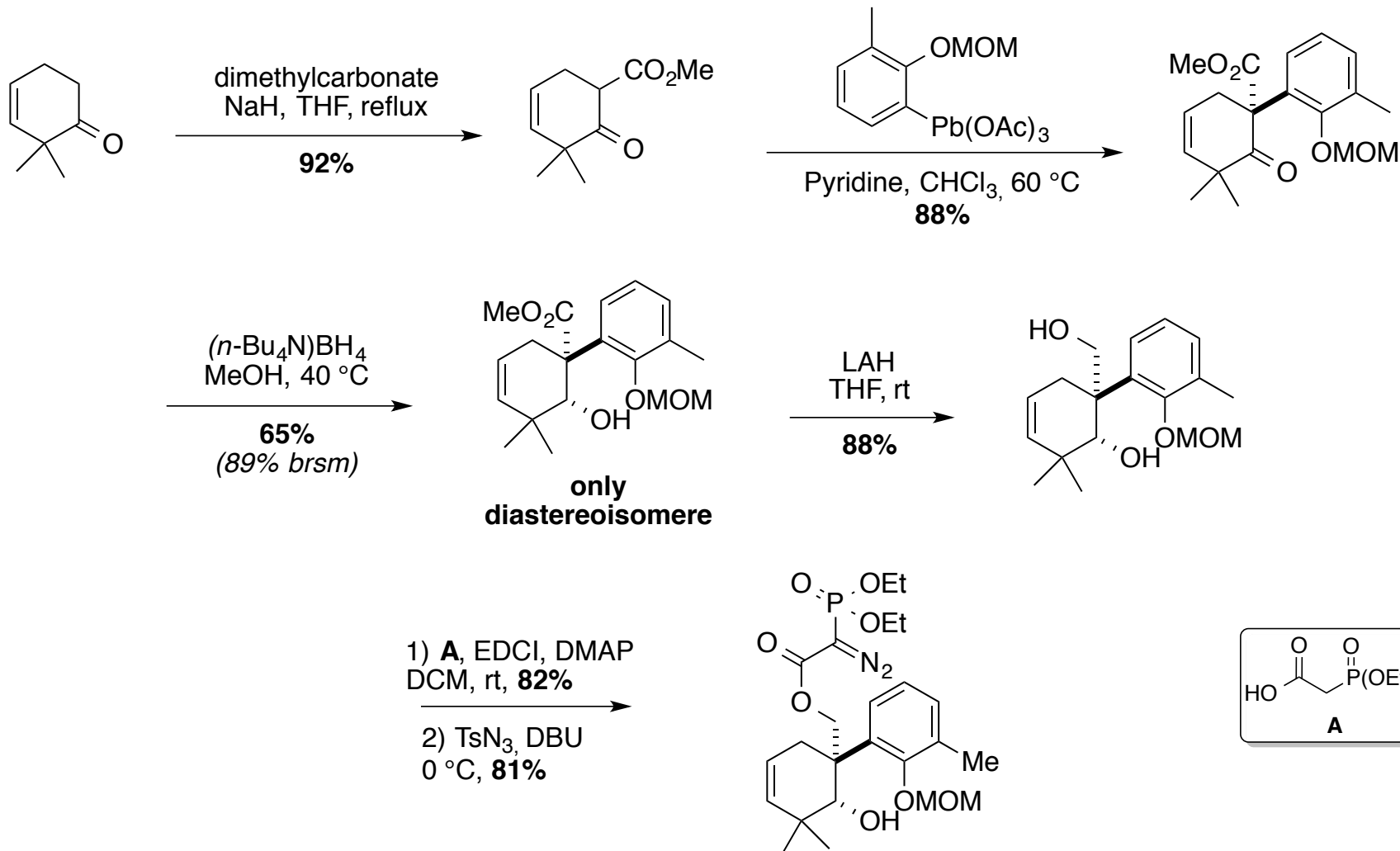
Conclusion

- > Only 2 enantioselectives total syntheses
 - Zakarian (2014): 26 Steps, Overall Yield = 0.2%
 - Thomson (2014): 18 Steps, Overall Yield = 1%

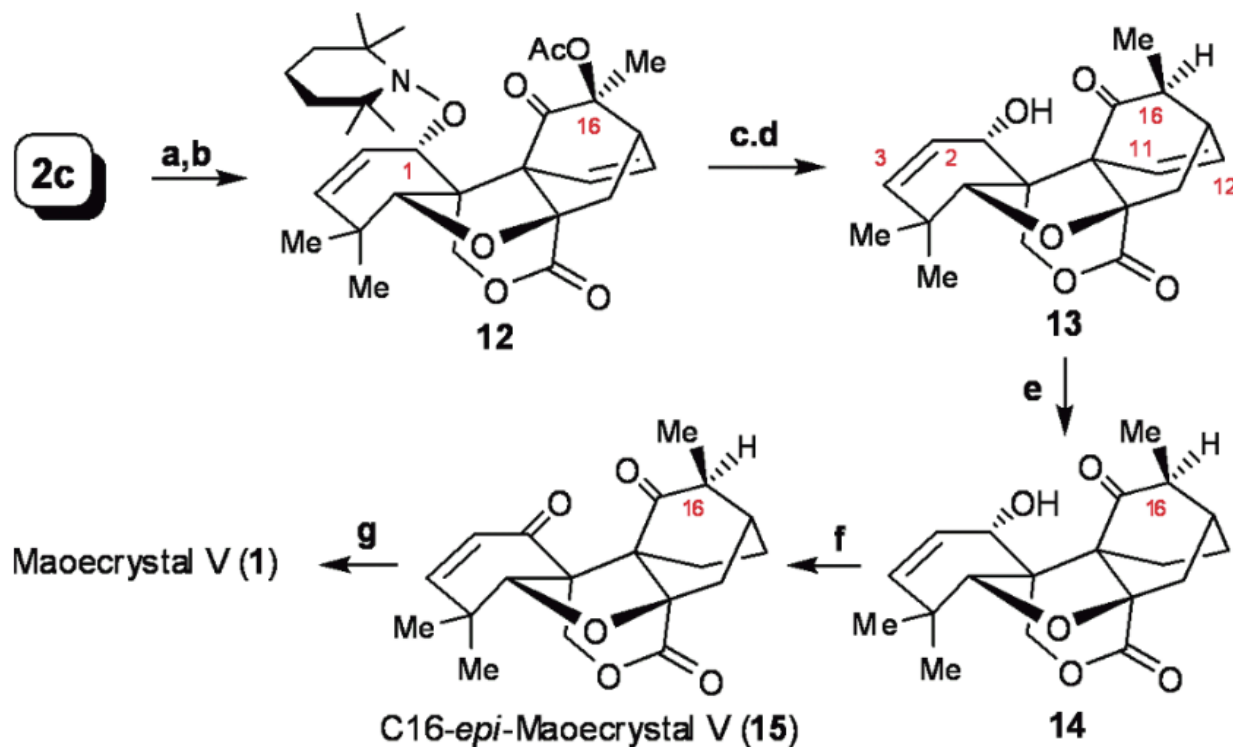
- > Key Steps:
 - Zakarian: Stereoselective Rh-catalyzed C-H functionalization and intramolecular Diels-Alder
 - Thomson: Heck spirocyclization, oxidative cyclodearomatization, intermolecular Diels-Alder

**Thank's
for
your attention**

Yang's Synthesis – Synthesis of the key intermediate

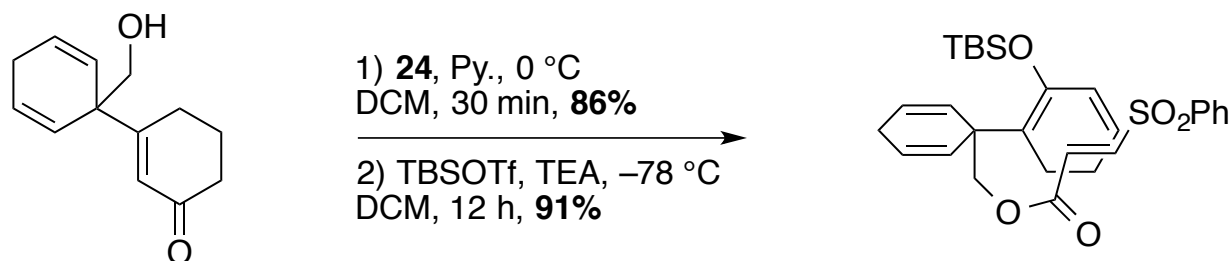
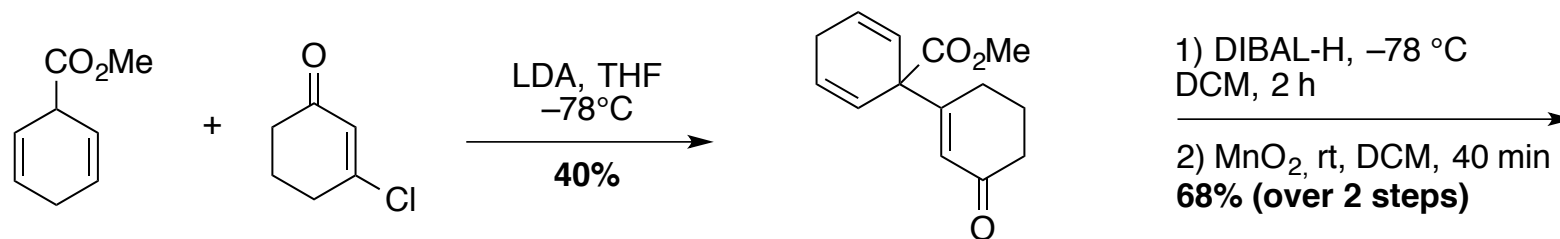


Yang's Synthesis – End of the Synthesis

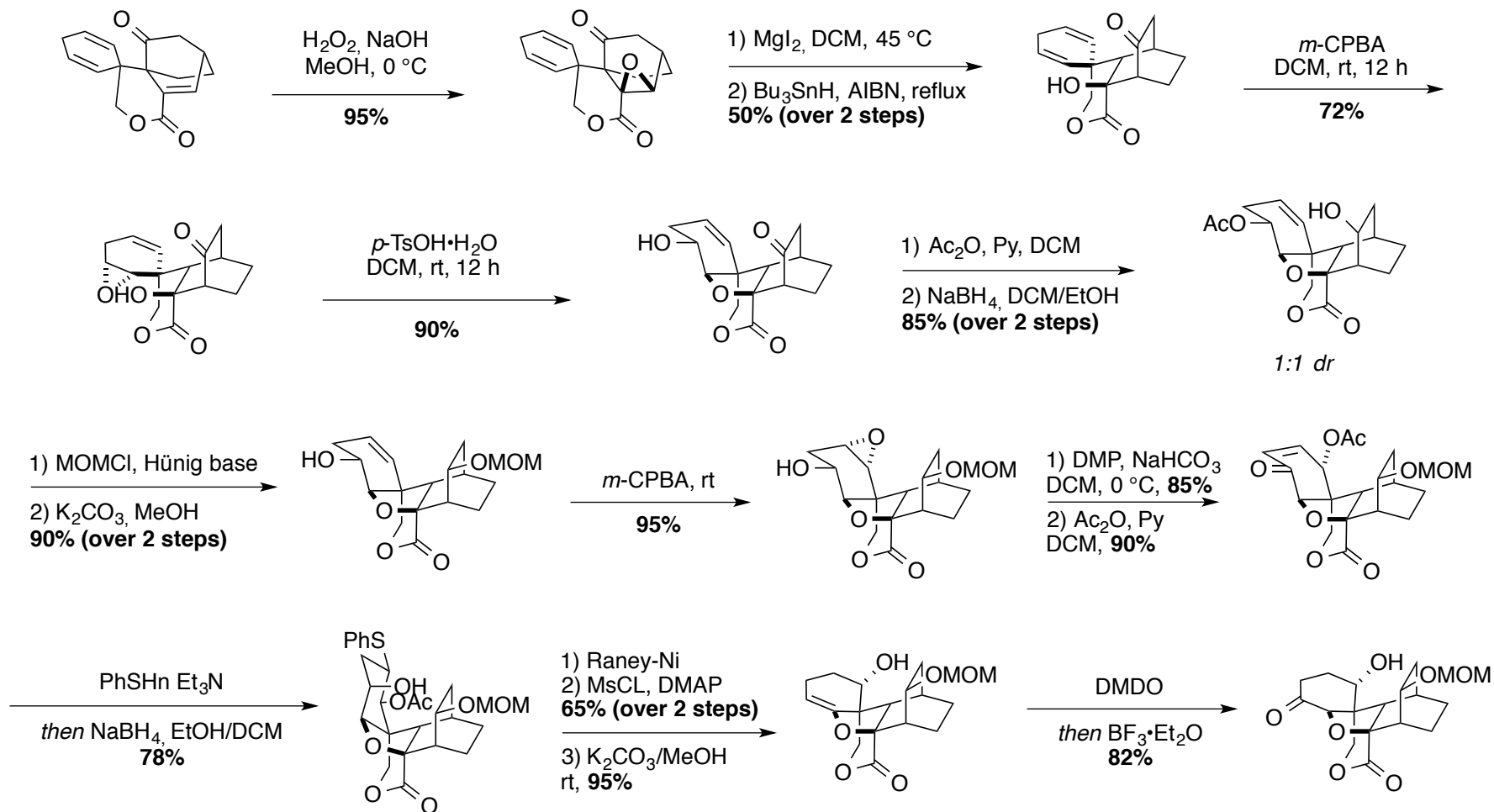


^a Reagent and conditions: (a) NBS, (PhCO₂)₂, CCl₄, reflux, 2 h, 90%; (b) Bu₃SnH, TEMPO, PhH, reflux, 2 h, 75%; (c) Zn, AcOH, THF, H₂O, 70 °C, 2 h, 85%; (d) SmI₂, THF, MeOH, rt, 10 min, 88%; (e) Lindlar cat. MeOH, THF, rt, 2 h, 92%; (f) DMP, CH₂Cl₂, rt, 1 h, 88%; (g) DBU, toluene, 100 °C, 1 h, 48% (90% brsm).

Danishefsky's Synthesis – Synthesis of the key intermediate



Danishefsky's Synthesis – End of the Synthesis



Danishefsky's Synthesis – End of the Synthesis

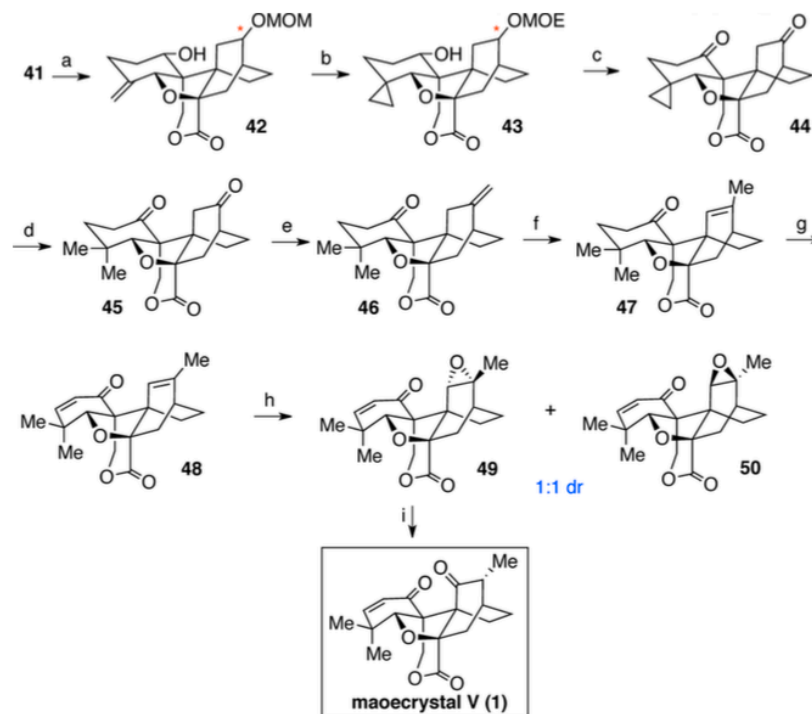
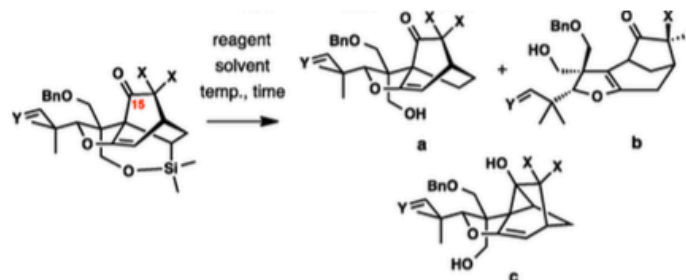


Figure 10. Completion of the synthesis of maoecrystal V (1). Key: (a) Lombardo reagent, DCM, room temperature, 85%; (b) CH_2I_2 , Zn/Ag, Et_2O , 36 °C, 88%; (c) PCC, DCM, room temperature, 76%; (d) H_2 , PtO_2 , AcOH, 40%; (e) Lombardo reagent, DCM, 0 °C, 80%; (f) *p*-TsOH· H_2O , benzene, 76 °C, 85%; (g) LDA, TMSCl, THF, -78 °C, 90%; then $\text{Pd}(\text{TFA})_2$, CH_3CN , 80%; (h) TFDO, CH_2Cl_2 , -78 °C → 0 °C, dr = 1:1, 90%; (i) $\text{BF}_3\cdot\text{OEt}_2$, DCM, room temperature, 85%.

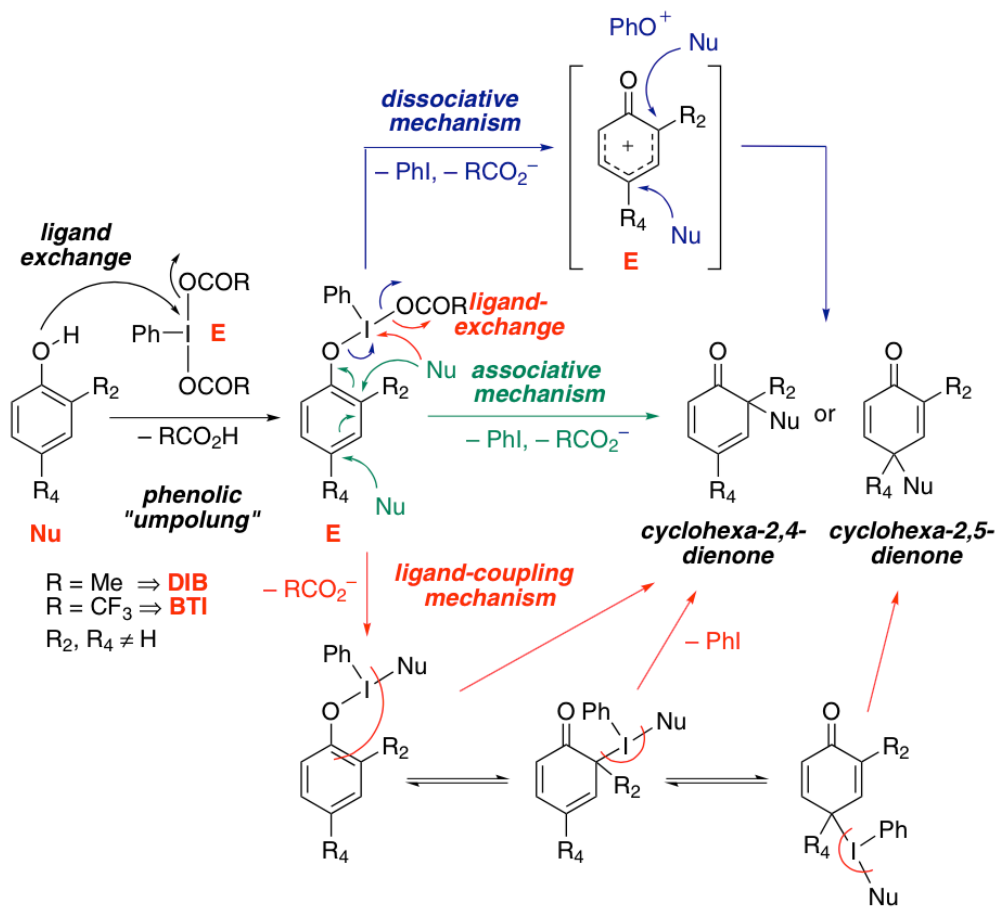
Zakarian's Synthesis

Table 4



entry	X	Y	reagent solvent	temp (°C), time (h)	a:b:c ^a (yield, a)
1	H	H, OBn	Bu ₄ NF, DMSO	0, 0.2	1.0:1.8:1 (20%)
2	H	H, OBn	Bu ₄ NF, DMSO	23, 0.2	0.9:1.5:1
3	H	H, OBn	Bu ₄ NF, DMF	-20, 2	1.5:1.3:1
4	H	H, OBn	Bu ₄ NF, DMF	0, 0.5	2.3:1.8:1
5	H	H, OBn	Bu ₄ NF, DMF	23, 0.3	1.9:1.6:1
6	H	H, OBn	Bu ₄ NF, DMF, (slow addition)	23, 3.2	1.6:1.3:1
7	H	H, OBn	Bu ₄ NF, NMP	0, 1.0	1.4:1.7:1
8	H	H, OBn	Bu ₄ NF, DMPU	0, 1.0	3.2:1.8:1
9	H	H, OBn	Bu ₄ NF, DMA	0, 0.5	2.4:1.7:1
10	H	H, OBn	Bu ₄ NF, DMSO- <i>t</i> -BuOH (4:1)	75, 1	0.3:1.2:1
11	H	H, OBn	Bu ₄ NF, AcOH, THF	0, 1.0	no reaction
12	H	H, OBn	CsF, <i>t</i> -BuOH	70, 15	no reaction
13	H	H, OBn	Bu ₄ NPh ₃ SiF ₂ , DMSO	50, 36	0:20:1
14	H	H, OBn	Bu ₄ NPh ₃ SiF ₂ , <i>t</i> -BuOH	50, 15	no reaction
15	OE _t	CH ₂	Bu ₄ NF, DMF	23, 0.2	0.8:5.1:1 (7%)
16	OE _t	CH ₂	Bu ₄ NF, THF	23, 0.5	no reaction
17	OE _t	CH ₂	Bu ₄ NF, MeCN	23, 0.2	0:3.4:1
18	H	CH ₂	Bu ₄ NF, DMF	23, 0.5	1.1:3.6:1
19	H	H, OPMB	Bu ₄ NF, DMA	0, 1.0	2.9:2.2:1
20	H	H, OPMB	Bu ₄ NF, DMPU	0, 1.0	3.2:1.9:1 40–50%

Oxidative aromatization

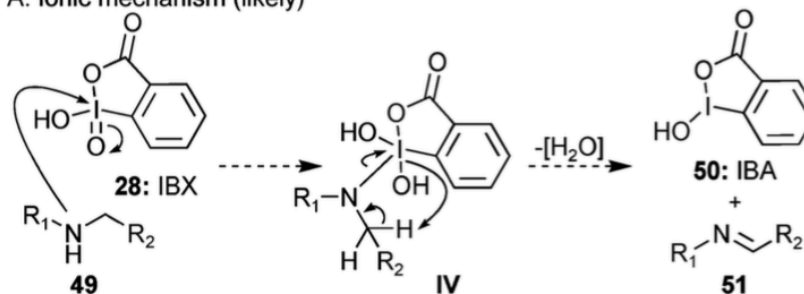


Scheme 4.

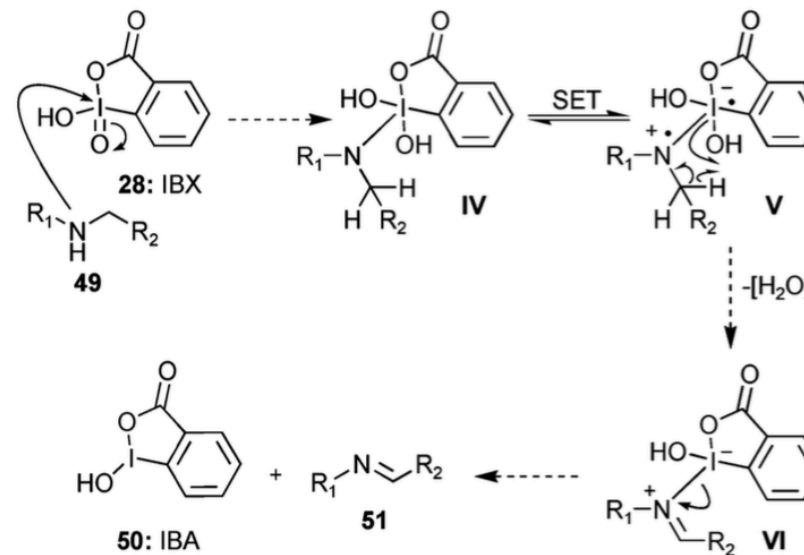
IBX oxidation

Scheme 3. Proposed (A) Ionic and (B) Single Electron Transfer (SET) Mechanisms for the Oxidation of Amines Mediated by IBX

A: Ionic mechanism (likely)

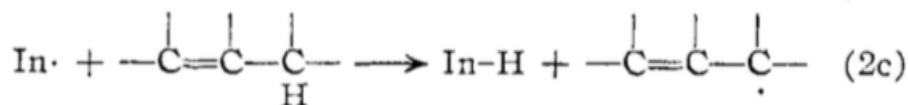
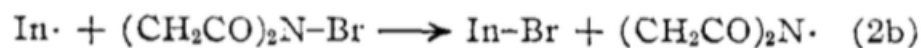
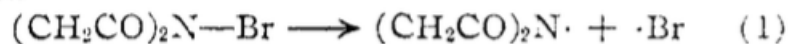


B: Single electron transfer (SET) mechanism (cannot be excluded)

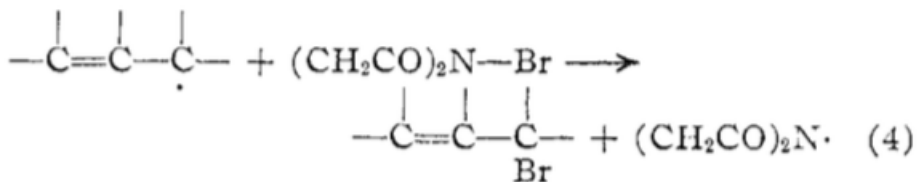
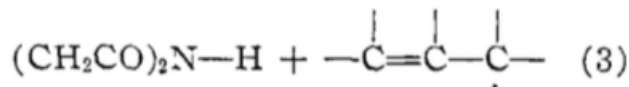
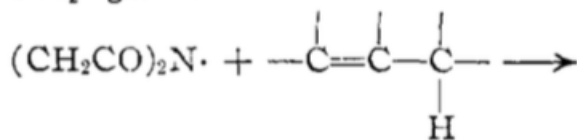


Allylic Bromination

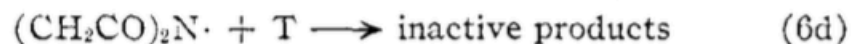
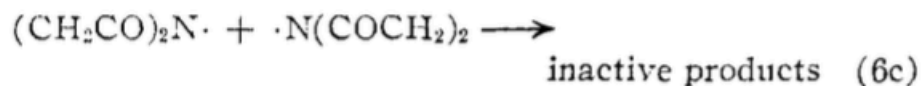
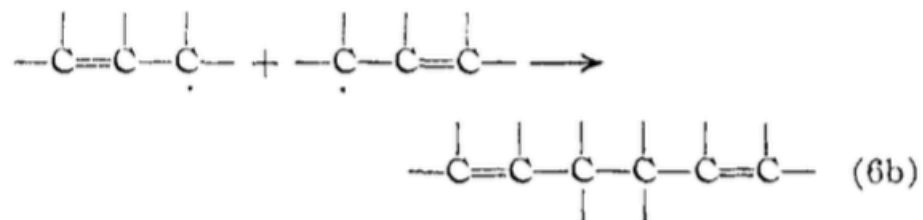
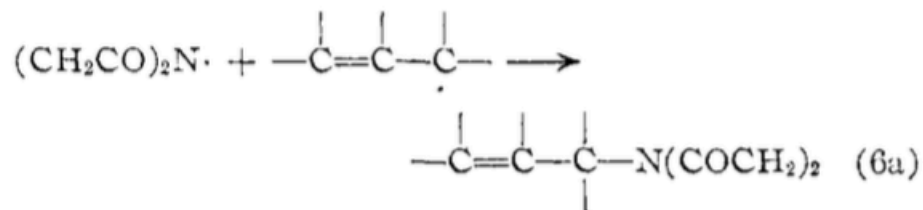
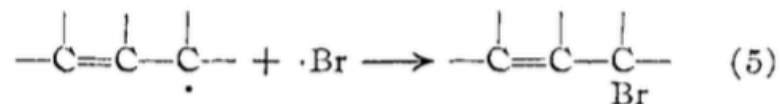
Initiation



Propagation

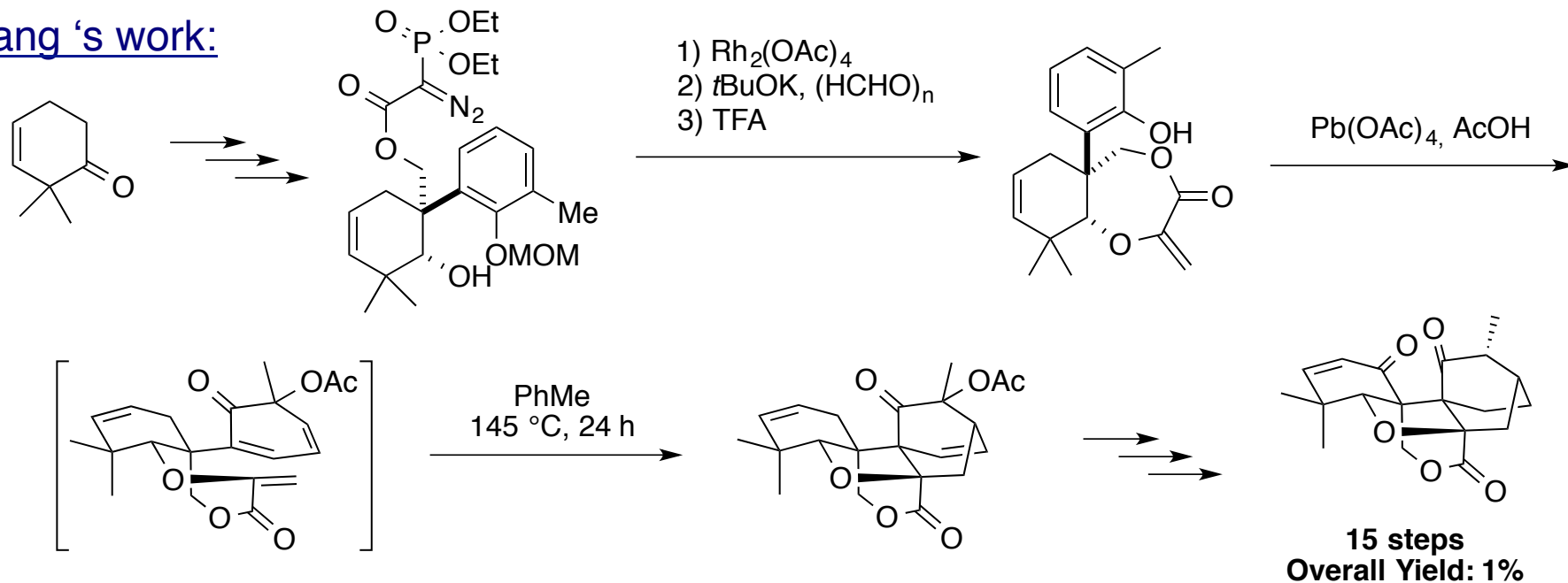


Termination

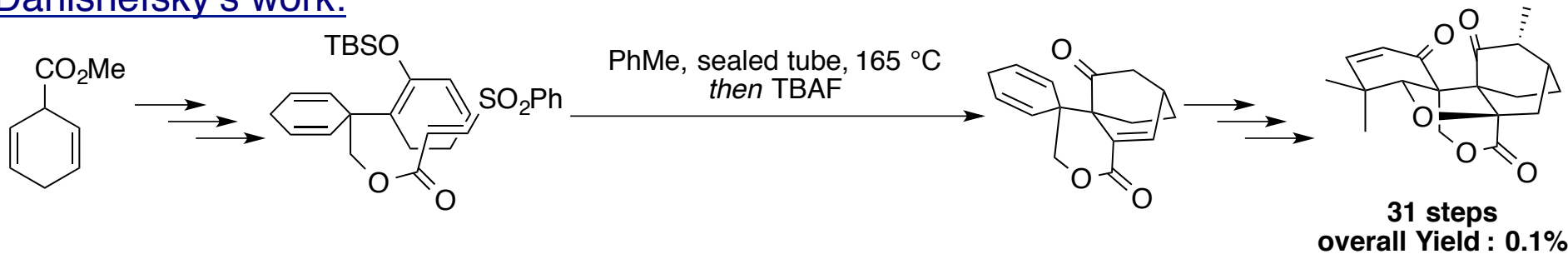


Previews Syntheses

Yang 's work:



Danishefsky's work:



Thomson's Strategy

