## Total Synthesis of (+)-Chinensiolide B

Problem:



- 1) Give the reagents for the transformation  ${\bf 1}$  to  ${\bf 3}$
- 2) What is the product A?
- 3) Give the product **B** and give a rational for the stereochemical outcome.
- 4) Explain the byproduct in the reaction from **B** to **4**. (Mechanism)
- 5) Give the reagent for the transformation of 5 to 6 as well as product 6.
- 6) Explain the reagents for the transformation of 6 to 7

Solution:



Rational for stereochemical outcome of reaction 3 to B:



The Felkin-Ahn model explains the remarkable diastereofacial selectivity on the aldehyde whereas the Zimmerman-Traxler model explains the *trans* diastereoselectivity of the allylboronation step.



Mechanism from **B** to **4** (Grieco Elimination):

 $o-NO_{2}-C_{6}H_{4}SeCN + PBu_{3} \longrightarrow o-NO_{2}-C_{6}H_{4}SePBu_{3}CN$   $e^{\oplus}O-NO_{2}-C_{6}H_{4}SePBu_{3}CN + RCH_{2}OH \longrightarrow o-NO_{2}-C_{6}H_{4}Se^{\oplus} + HCN + RCH_{2}OPBu_{3}$   $o-NO_{2}-C_{6}H_{4}Se^{\oplus} + RCH_{2}OPBu_{3} \longrightarrow o-NO_{2}-C_{6}H_{4}SeCH_{2}R + O=PBu_{3}$   $o-NO_{2}-C_{6}H_{4}SeCH_{2}R + H_{2}O_{2} \longrightarrow (NO_{2}-C_{6}H_{4}SeCH_{2}R + O=PBu_{3})$ 

Byproduct is formed by conjugate addition of the cyanide anion to the enolate.

Choice of reagents for the transformation of 6 to 7:

By the use of LAH, the lactone was over-reduced to give the fully saturated triol and the LiEt<sub>3</sub>BH underwent preferentially conjugate addition on the  $\alpha$ -methylene group then opening the epoxide.

Therefore, the lactone was reduced by DIBALH to the diol, followed by regionand chemoselective epoxide opening with LiEt<sub>3</sub>BH. Re-oxidation with MnO<sub>2</sub> reformed the  $\alpha$ -methylene  $\gamma$ -lactone.

## References:

T. G. Elford; D. G. Hall, *J. Am. Chem. Soc.* **2010**, *132*, 1488 *trans*-selectivity: J. W. J. Kennedy; D. G. Hall, *J. Org. Chem.* **2004**, *69*, 4412 Grieco Elimination: P.A. Grieco, S. Gilman, M. Nishizawa, *J. Org. Chem.* **1976**, *41*, 1485-1486

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