Synthesis of pleuromutilin antibiotics

1. How would you synthetize the compound 2 from 1? (with the relative stereochemistry shown)

c) TMSOTf (2 equiv), (TMSOCH₂)₂ (7 equiv), DCM, 20°C.

2. Starting from 2, provide the synthetic steps needed to obtain 3 (give the corresponding reagents and conditions)

d) tBuLi (4.4 equiv), 8 (2.4 equiv) Et₂O, -45°C, then 7 (1 equiv) -45°C -work up-, then HCl 1M, THF, 0°C.

- 3. Explain the stereoselectivity of the hydrocyanation reaction leading to compound **5**. What is the name of this reaction?
- 4. Give the reagents and conditions to perform the transformation of **6** into **7** in one single step. Provide the corresponding reaction mechanism.
- 5. Propose the reaction mechanisms involved in the formation of **9** and **11**.

Bonus: How would you prepare the iodide 8 from the chiral precursor 12?

Synthesis of pleuromutilin antibiotics

Submitted by Camilo M.

Solution

1. How would you synthetize the compound **2** from **1?** (with the relative stereochemistry shown)

2. <u>Starting from **2**</u>, provide the synthetic steps needed to obtain **3** (give the corresponding reagents and conditions)

3. <u>Explain the stereoselectivity of the hydrocyanation reaction leading to compound **5**. What is the name of this reaction?</u>

Nagata hydrocyanation: Coordination of aluminum with the ester moiety directs the nucleophilic attack of CN⁻. Once the conjugated addition takes place, the so generated enolate can coordinate with aluminum as well. By doing this, steric hindrance favors the protonation *anti* to the CN group, leading to a trans connectivity of the two rings. Nevertheless, the corresponding epimer formed (selectivity 3:1) is selectively reduced by DIBAL and separated from the original mixture. At the end, by adding base, the product isomerizes to the *cis* product.

4. <u>Give the reagents and conditions to perform the transformation of 6 into 7 in one single step.</u>

Provide the corresponding reaction mechanism

5. <u>Propose the reaction mechanisms involved in the formation of **9** and **11**.</u>

FOR 9

$$R \cap I \xrightarrow{tBuLi} R \cap Li$$

FOR **11**

$$\begin{array}{c|c}
Sml_2 \\
SET
\end{array}$$

$$\begin{array}{c}
Sml_2 \\
SET
\end{array}$$

$$\begin{array}{c}
OSml_2 \\
OSml_2 \\
OSml_2
\end{array}$$

In terms of stereoselectivity, the protonation taking place during the keto-enol tautomerism is directed by the bulkier vinyl group (relative to the methyl substituent)

Bulkier vinyl group directs the protonation (syn to the methyl group)

Bonus: How would you prepare the iodide **8** from the chiral precursor **12**?

O O O OPMB

CH₃

THF,
$$-78 \rightarrow 0$$
 °C

19

NaHMDS, PMBOCH₂CI

O O OPMB

CH₃

THF, $-78 \rightarrow 0$ °C

20