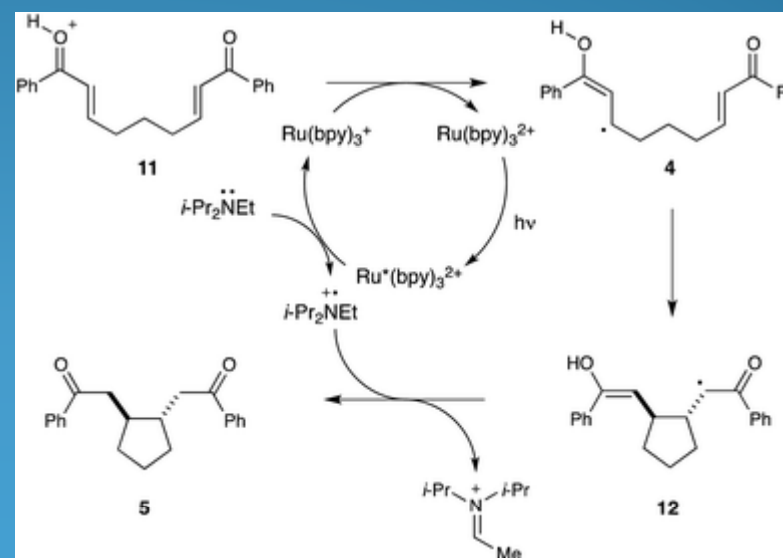
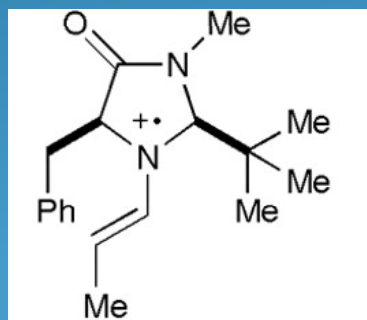
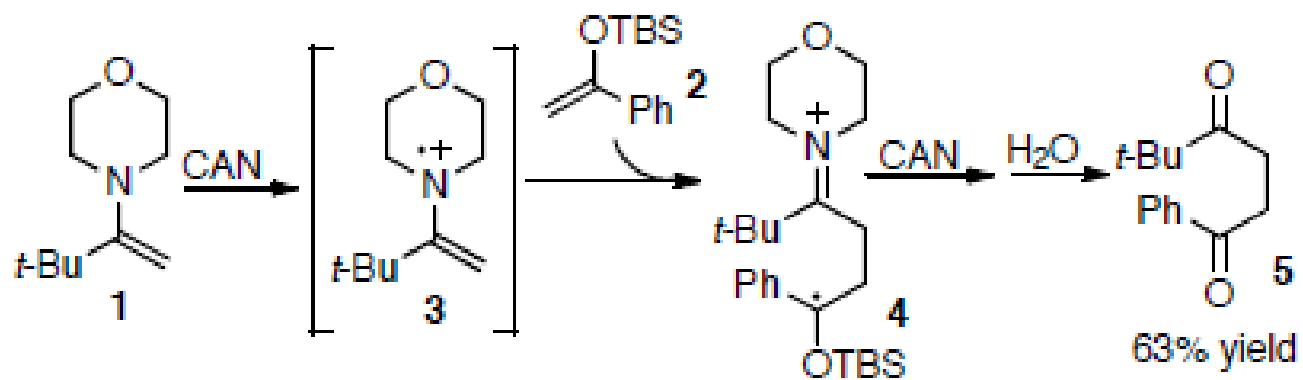


Radicals in Organocatalysis and Photocatalysis

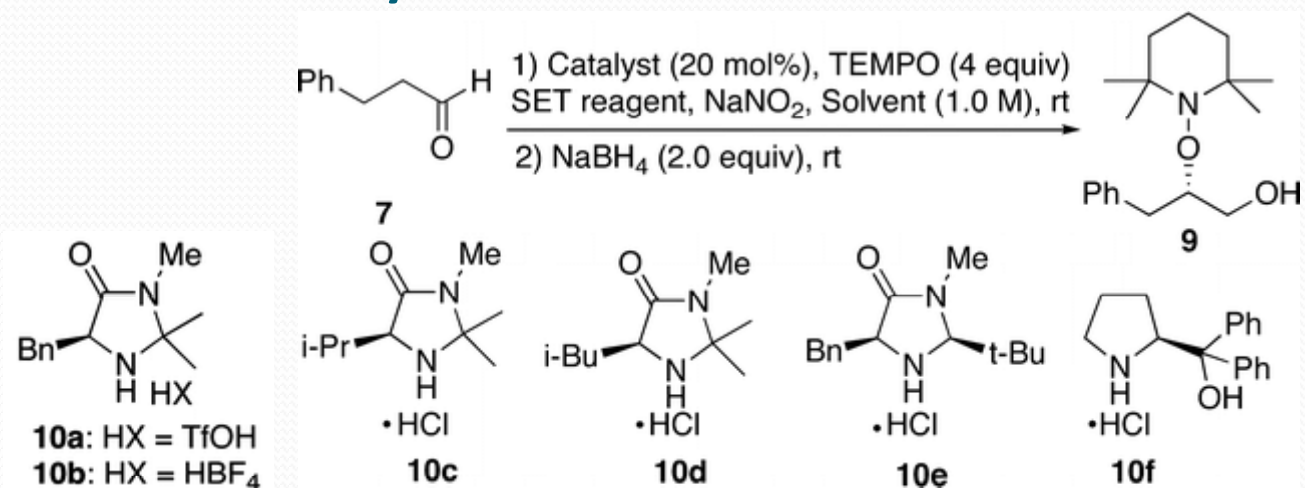


SOMO activation



Narasaka, K.; Okauchi, T.; Tanaka, K.; Murakami, M. *Chem. Lett. (Jpn.)* **1992**, *21*, 2099

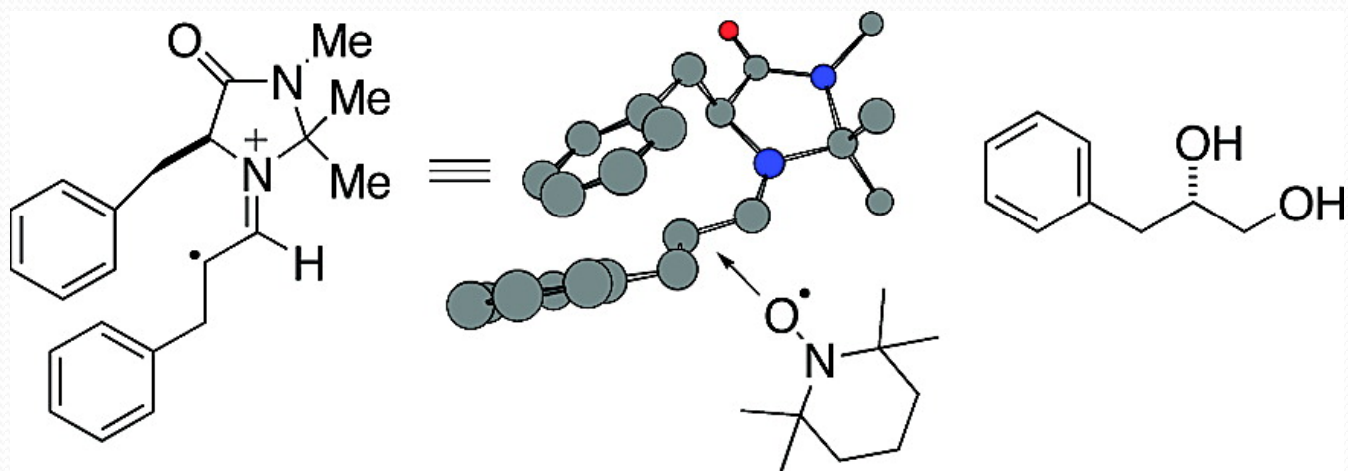
SOMO catalysis



entry	SET reagent (mol %)	ligand	NaNO ₂ (equiv.)	solvent	Yield (%)	ee (%)
1	FeCl ₃ (10)	10b	0.3	DMF	83	72
2	FeCl ₃ (10)	10c	0.3	DMF	75	5
3	FeCl ₃ (10)	10d	0.3	DMF	64	46
4	FeCl ₃ (10)	10e	0.3	DMF	26	0
5	FeCl ₃ (10)	10f	0.3	DMF	33	17

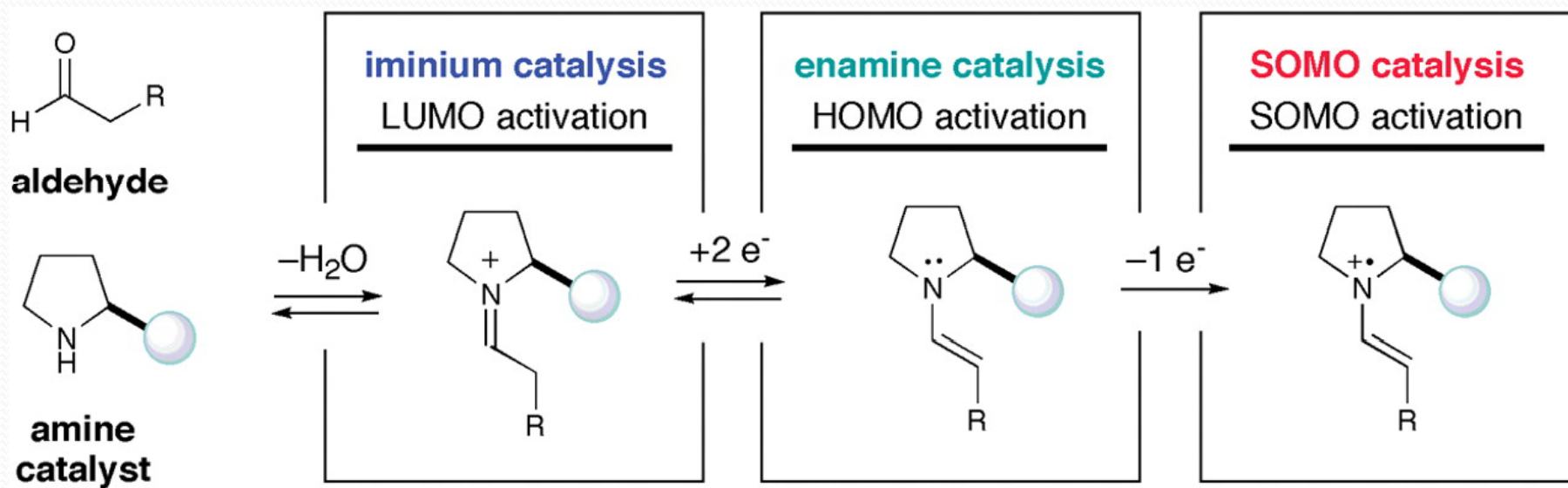
Sibi, M. P.; Hasegawa, M. *J. Am. Chem. Soc.* **2007**, *129*, 4124

SOMO catalysis



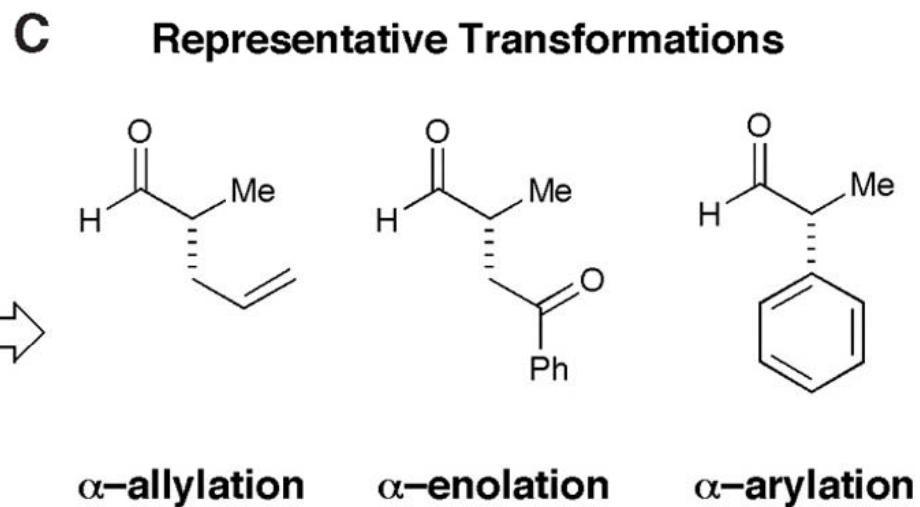
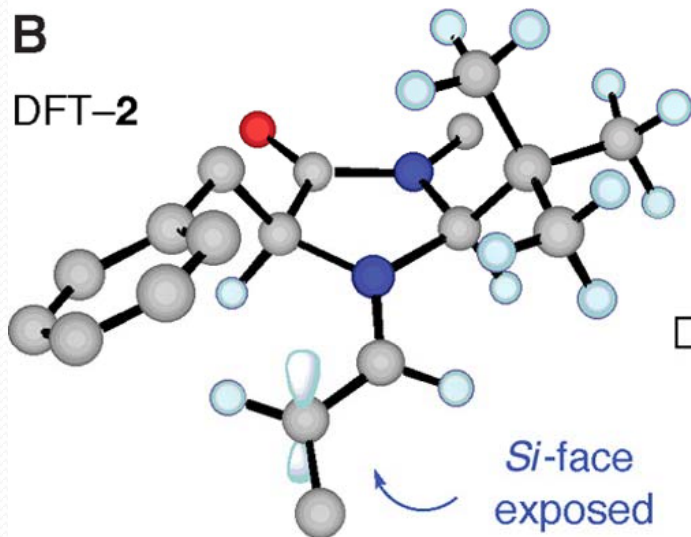
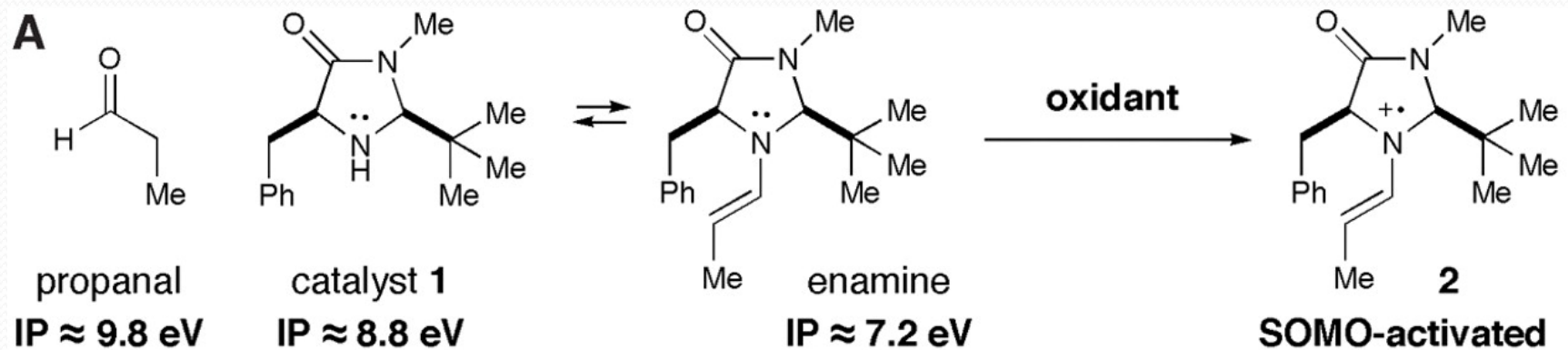
Sibi, M. P.; Hasegawa, M. *J. Am. Chem. Soc.* **2007**, *129*, 4124

SOMO catalysis

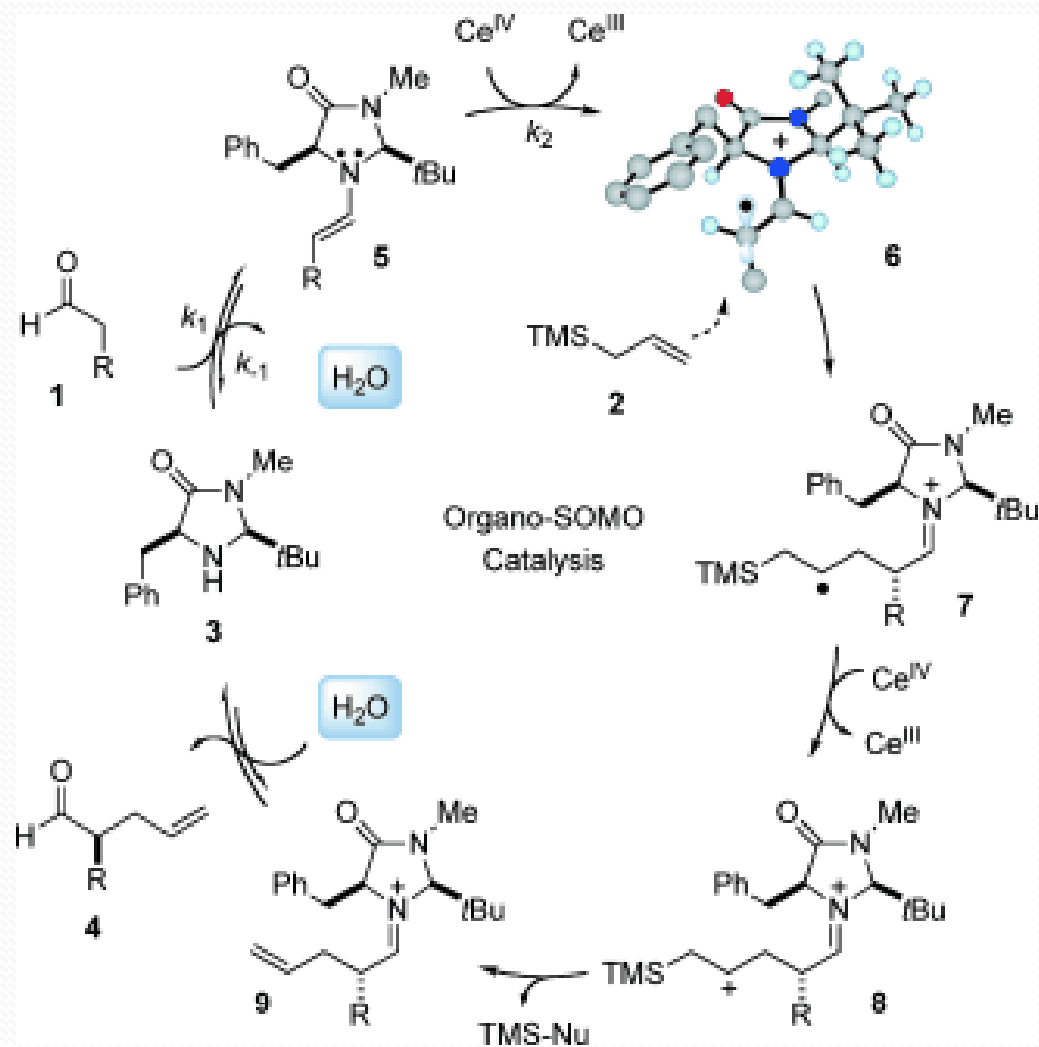


Beeson, T.D.; Mastracchio, A.; Hong, J.-B.; Ashton, K.; MacMillan, D.W.C. *Science* **2007**, *316*, 582

SOMO catalysis

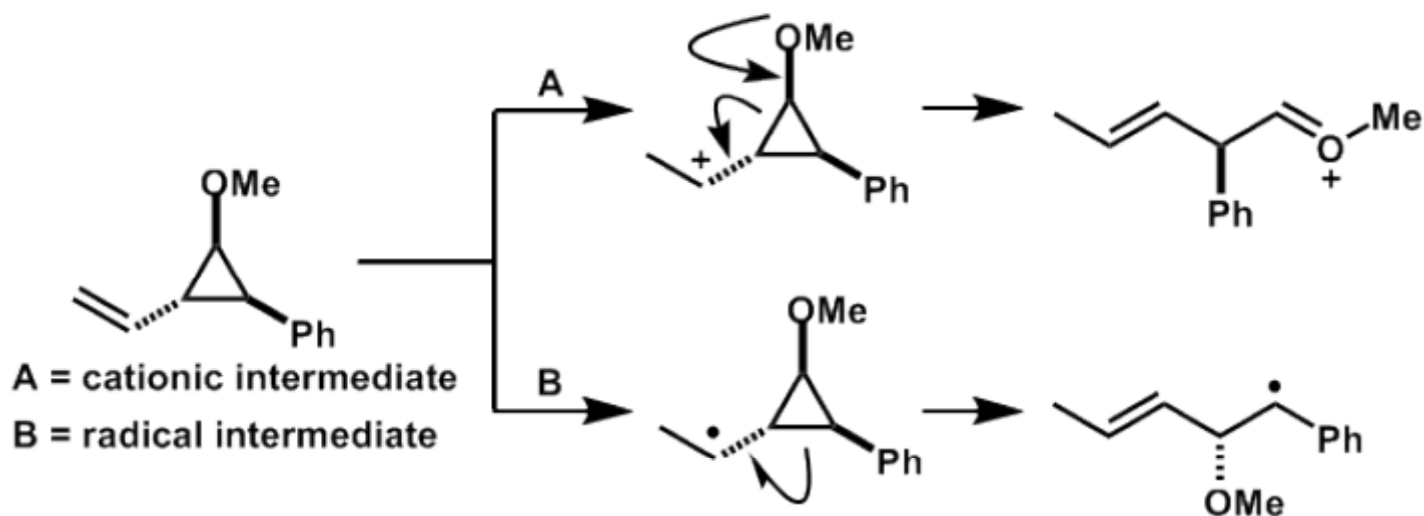
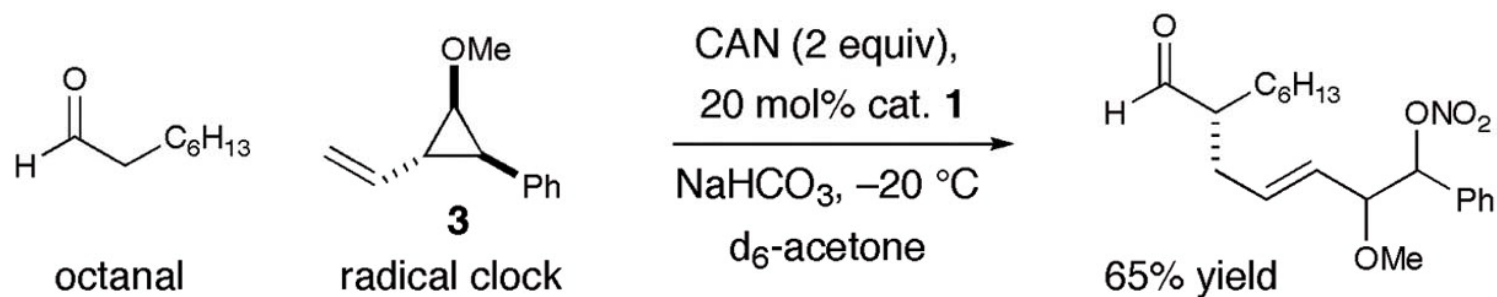


SOMO catalysis



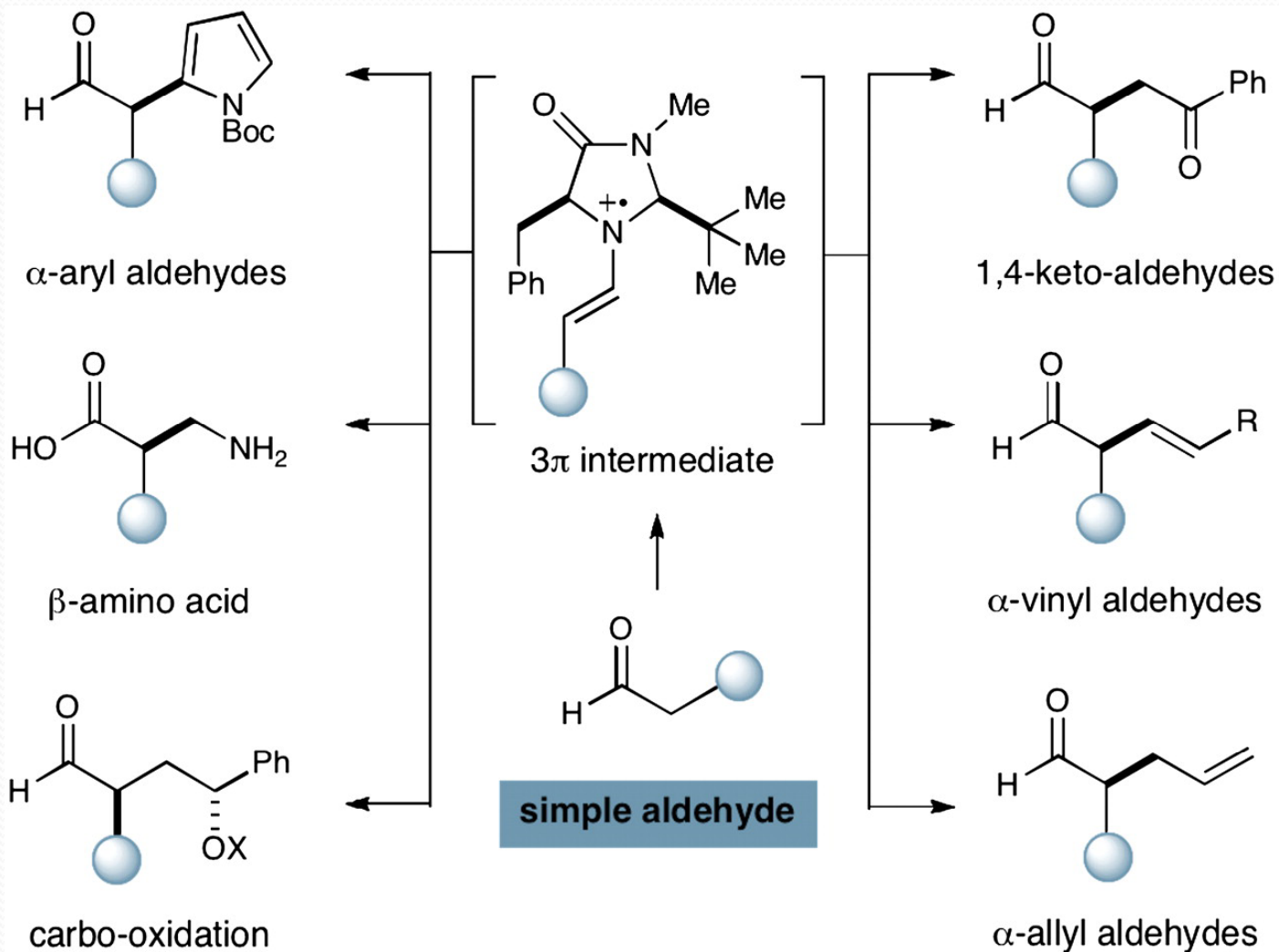
Devery, J.J., Conrad, J.C., MacMillan, D.W.C.; Flowers, R.A. *Angew. Chem. Int. Ed.* **2010**, *49*: 6106

SOMO catalysis

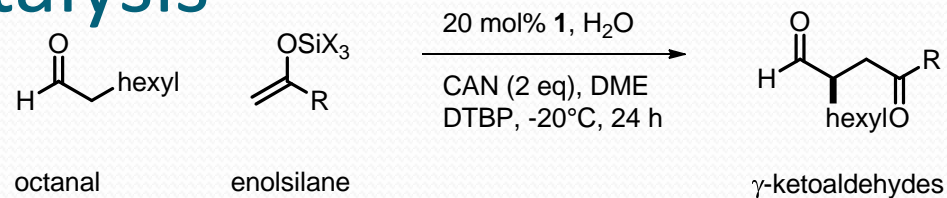


Beeson, T.D.; Mastracchio, A.; Hong, J.-B.; Ashton, K.; MacMillan, D.W.C. *Science*. 2007, 316, 582

SOMO catalysis



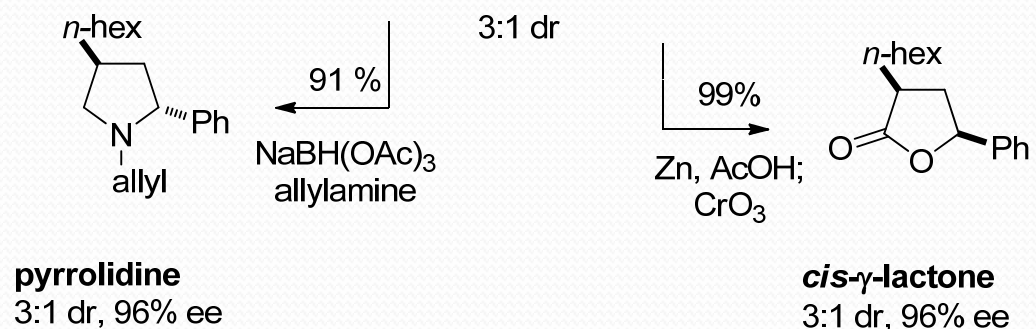
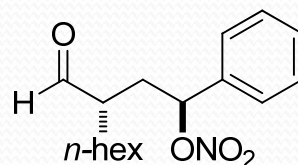
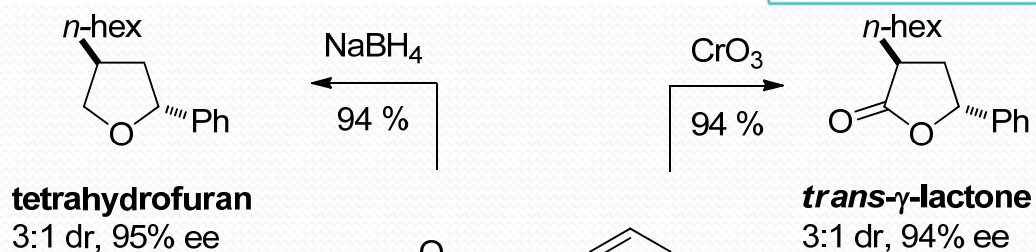
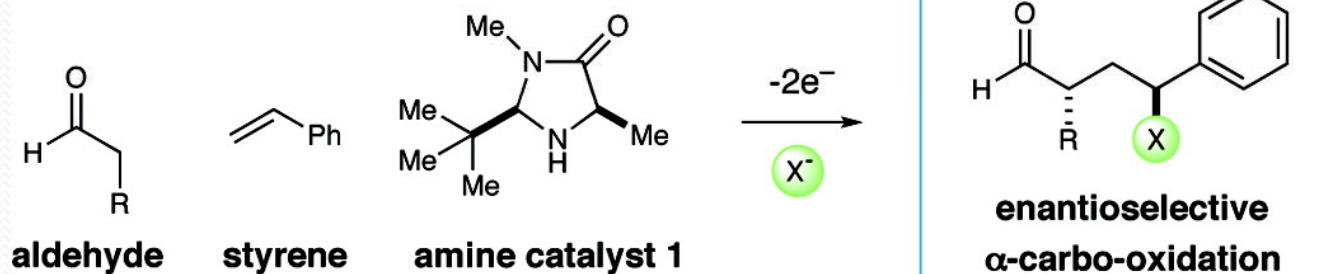
SOMO catalysis



entry	enolsilane	product	% yield	% ee
1			85	90
2			77	92
3			71	92
4			74	96
5			67	86
6			55	92

Jang, H.-Y.; Hong, J.-B.; MacMillan, D.W.C. *J. Am. Chem. Soc.* **2007**, *129*, 7004

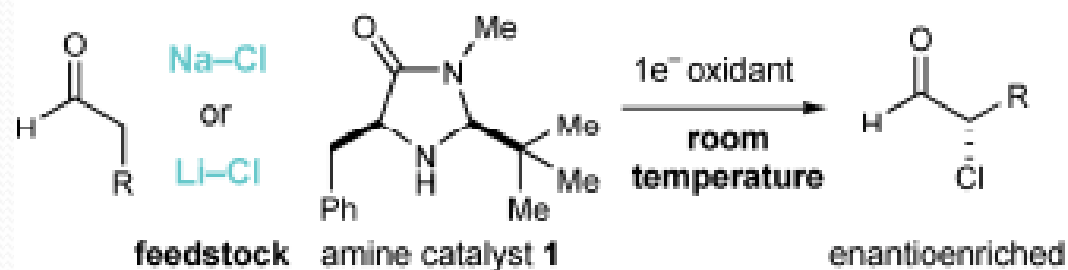
SOMO catalysis



Graham, T. H.; Jones, C. M.; Jui, N. T.; MacMillan, D.W.C. *J. Am. Chem. Soc.* **2008**, *130*, 16494

SOMO catalysis

SOMO Catalysis: Ambient Temp., Inexpensive Chlorine Source



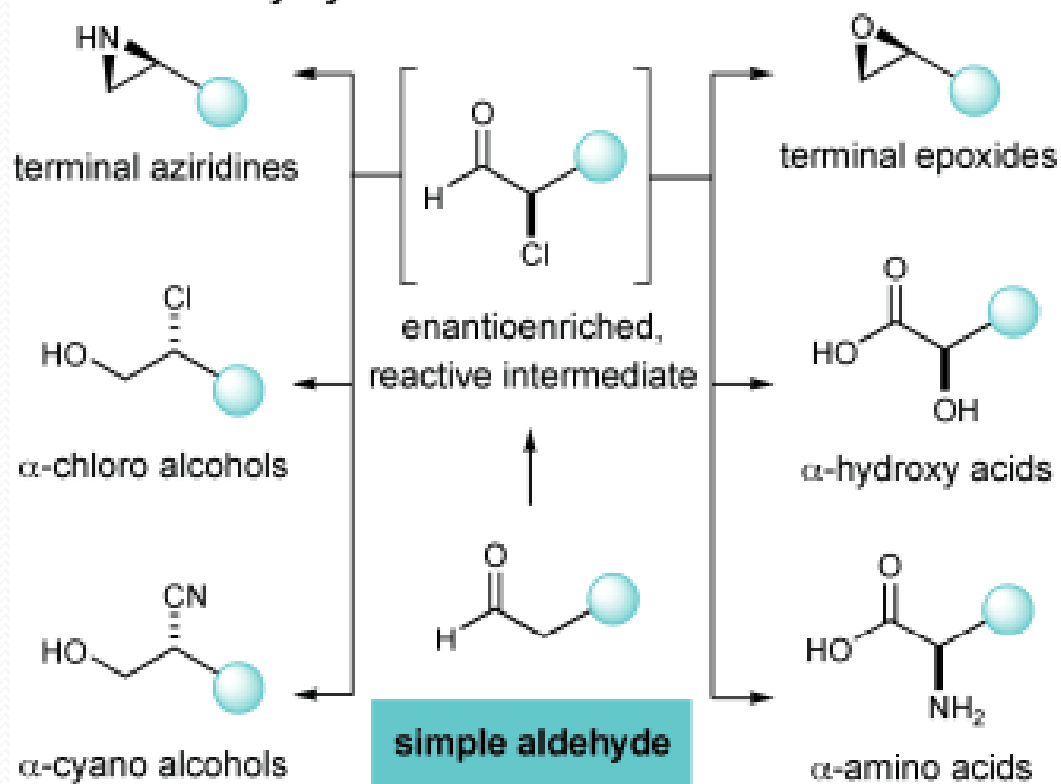
R = *n*-Hex (1.5 equiv), LiCl (2.2 equiv), Ce(NH₄)₂(NO₃)₆, catalyst 1, NaHCO₃

-40 °C = 95% yield, 92% ee

23 °C = 27% yield, 47% ee

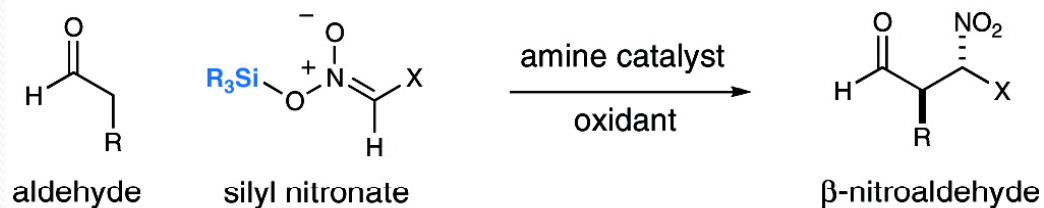
SOMO catalysis

Linchpin Catalysis via α -Chloroaldehydes: Access to Many Synthons

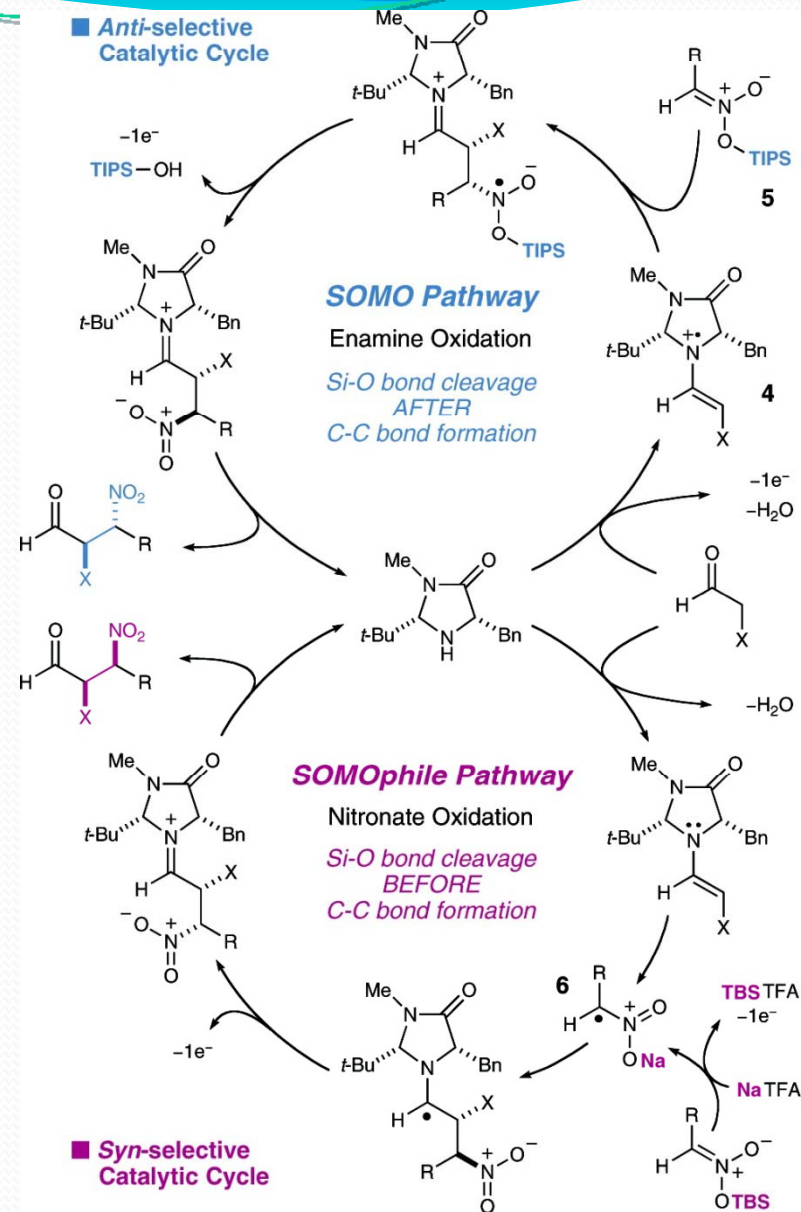


SOMO catalysis

Oxidative Organocatalytic Enantioselective Aldehyde Nitroalkylation

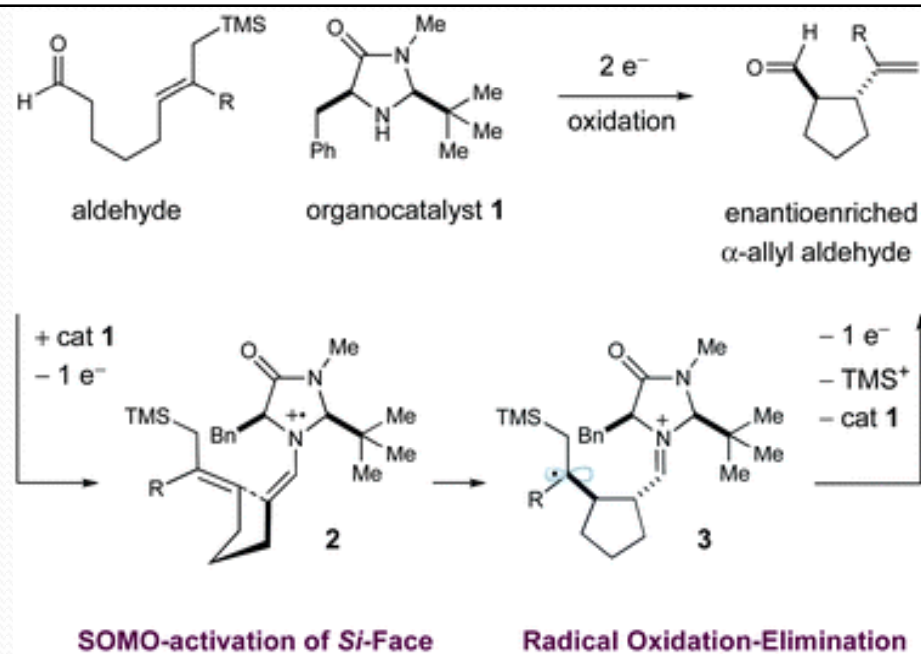
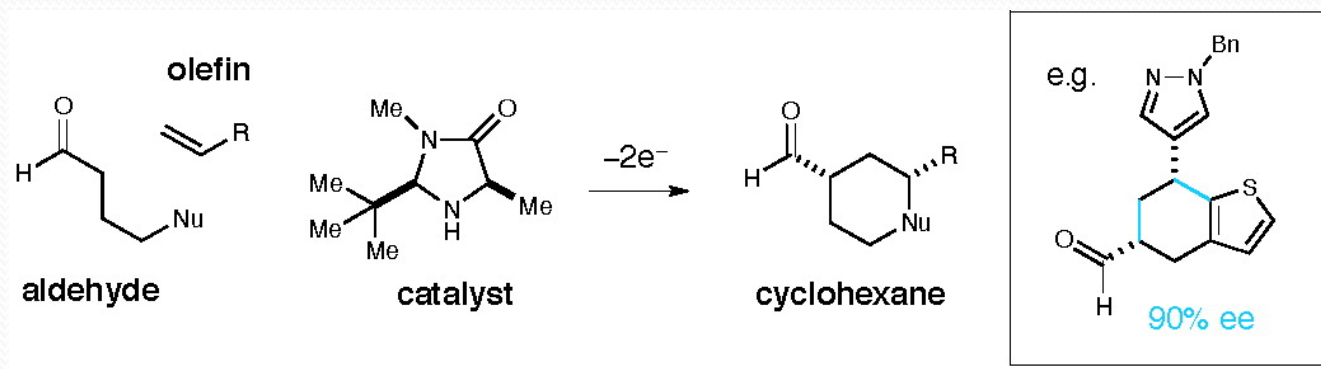


- **SOMO Catalysis or Enamine Catalysis**
- **Mechanism Controlled by Silyl Group**



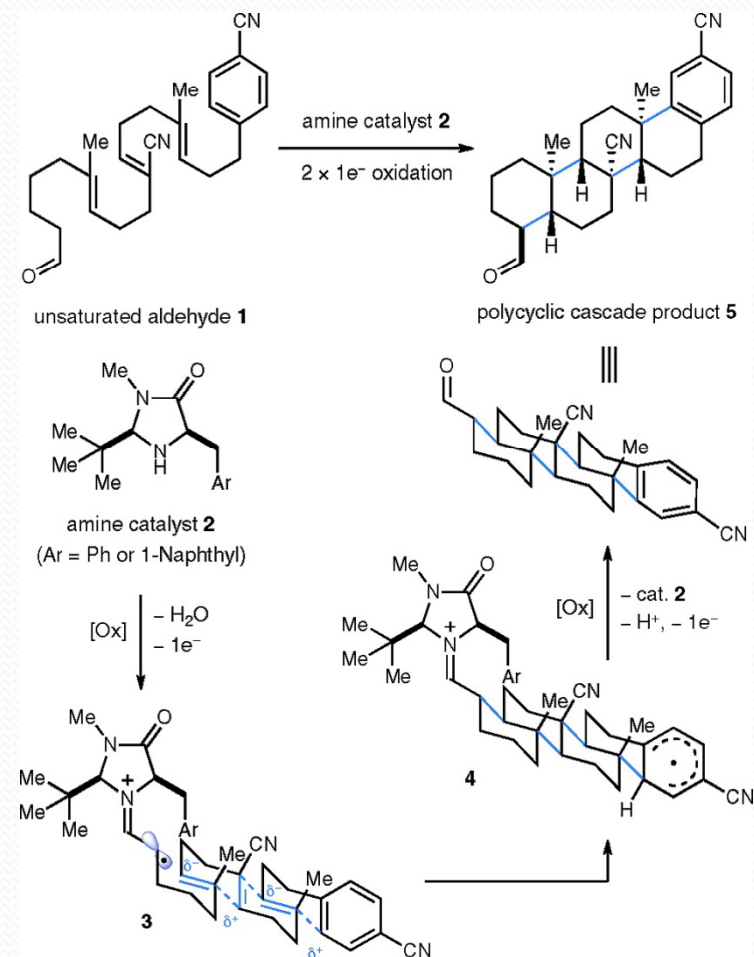
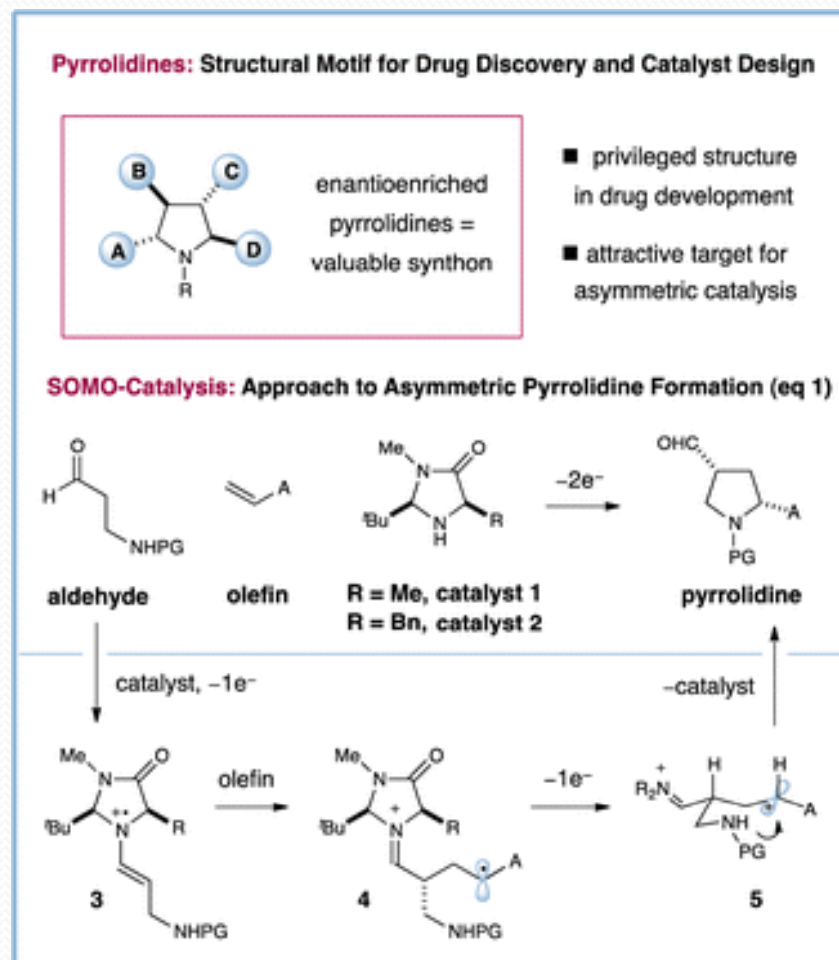
Wilson, J. E.; Casarez, A. D.; MacMillan, D.W.C. *J. Am. Chem. Soc.* **2009**, *131*, 11332

SOMO catalysis



Jui, N. T.; Lee, E. C. Y.; MacMillan, D.W.C. *J. Am. Chem. Soc.* **2010**, *132*, 10015
Pham, P. V.; Ashton, K.; MacMillan, D.W.C. *Chem. Sci.* **2011**, *2*, 1470

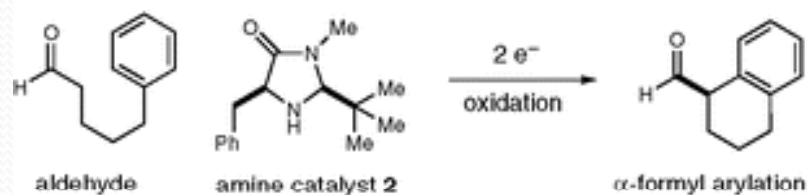
SOMO catalysis



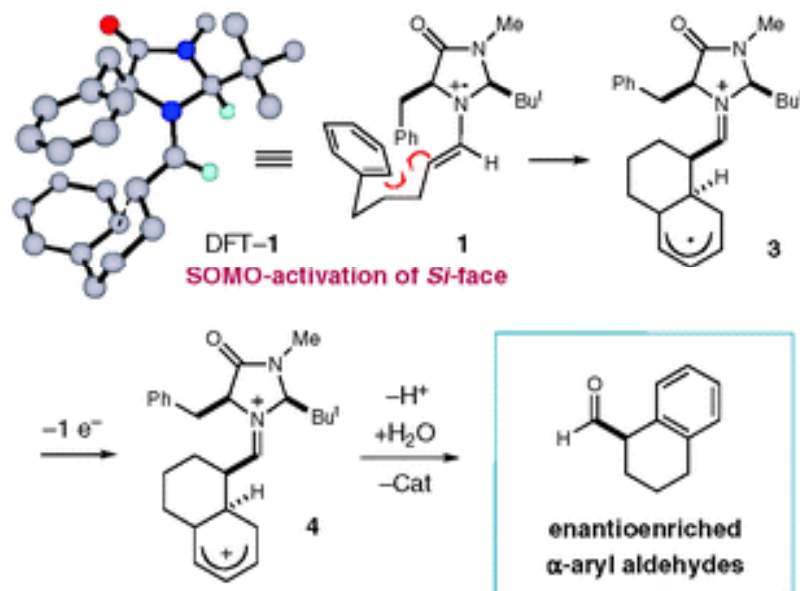
Jui, N. T.; Garber, J. A. O.; Finelli, F. G.; MacMillan, D.W.C. *J. Am. Chem. Soc.* **2012**, *134*, 11400
 Rendler, S.; MacMillan, D.W.C. *J. Am. Chem. Soc.* **2010**, *132*, 5027

SOMO catalysis

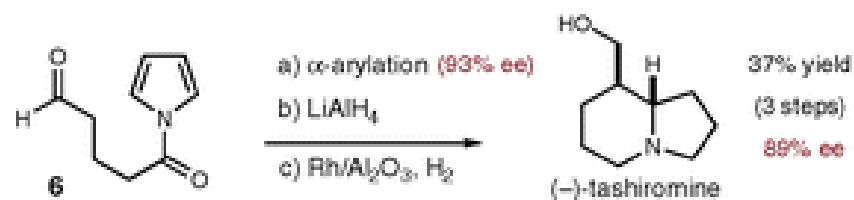
Enantioselective α -Arylation of Aldehydes via SOMO Catalysis (eq 1)



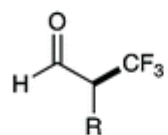
Mechanism of Organo-SOMO Catalyzed Aldehyde α -Arylation (eq 2)



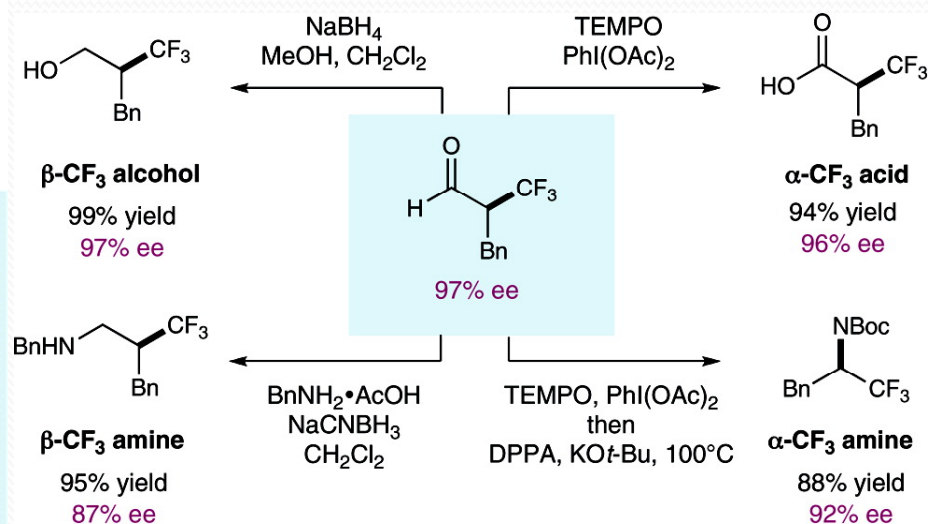
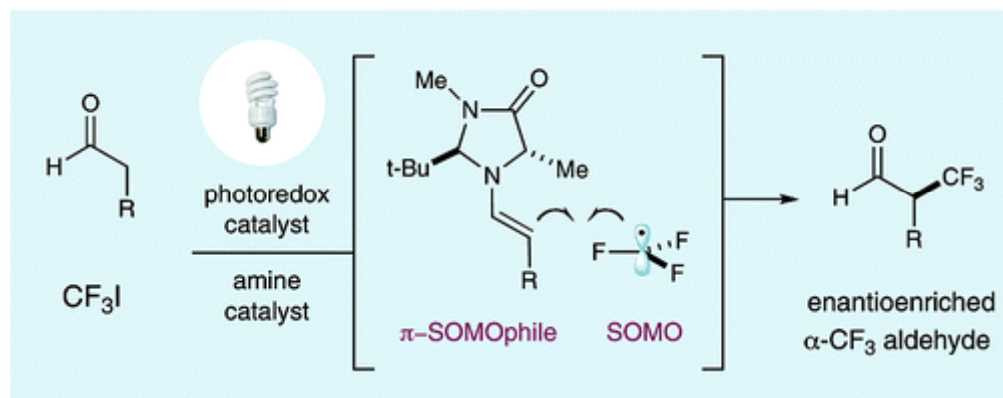
Enantioselective Synthesis of (-)-Tashiromine via Formyl α -Arylation



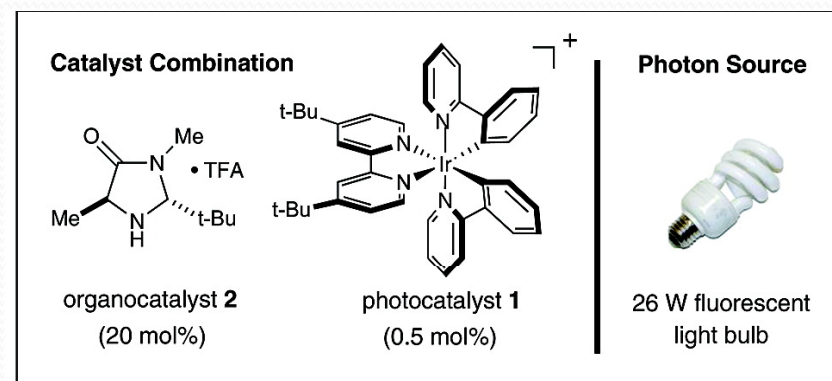
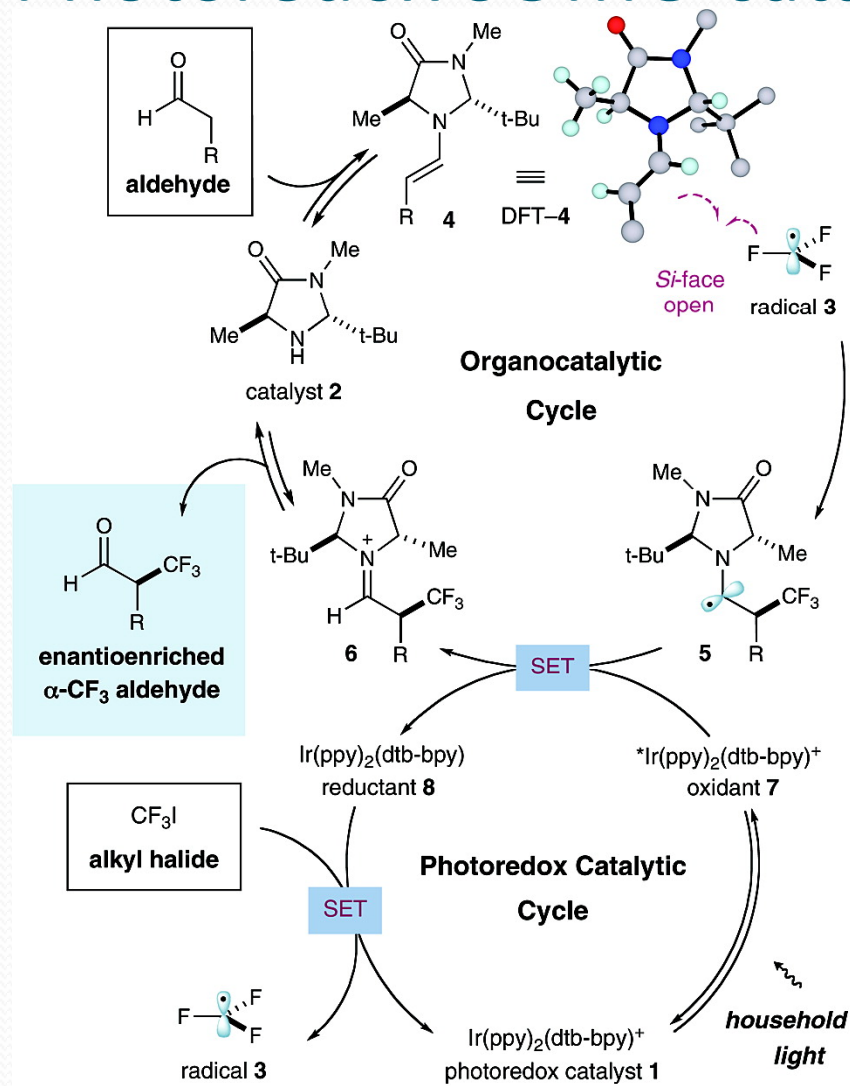
Photoredox SOMO catalysis



- Versatile synthon for medicinal agent synthesis
- Allows trivial access to enantioenriched CF₃-containing building blocks
- No known catalytic routes to α -formyl CF₃

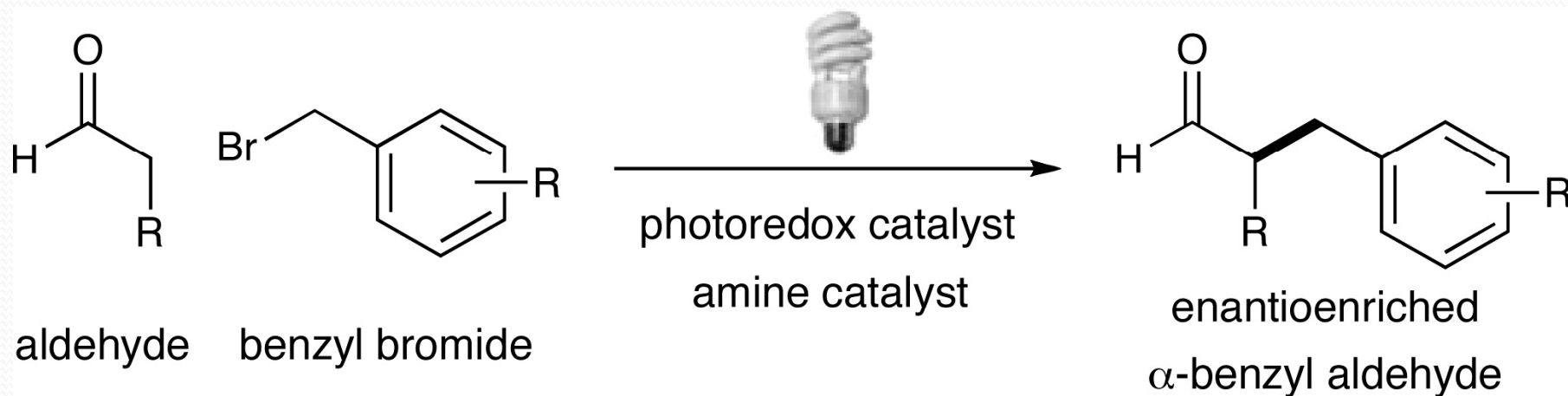


Photoredox SOMO catalysis



Nagib, D. A.; Scott, M. E.; MacMillan, D.W.C. *J. Am. Chem. Soc.* **2009**, *131*, 10875

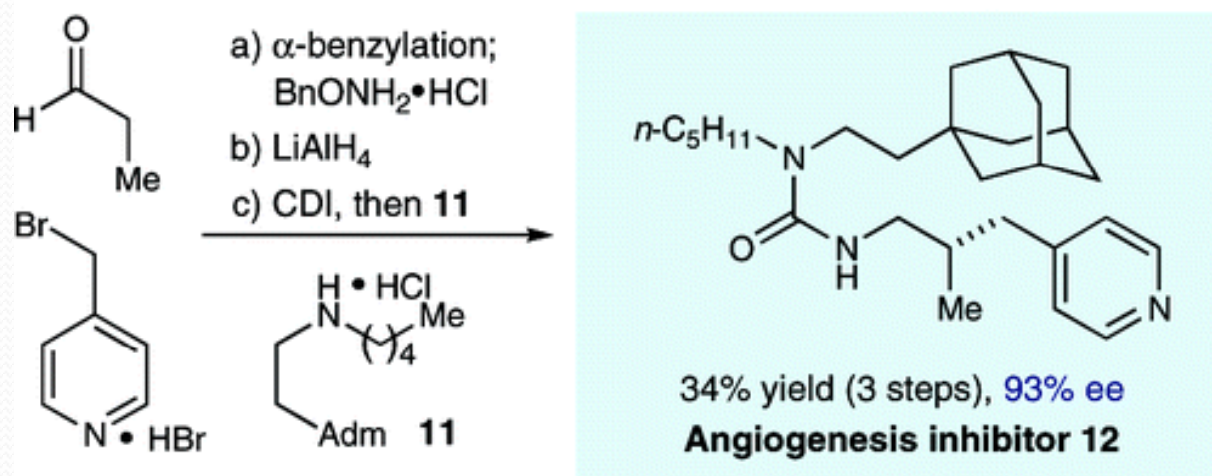
Photoredox SOMO catalysis



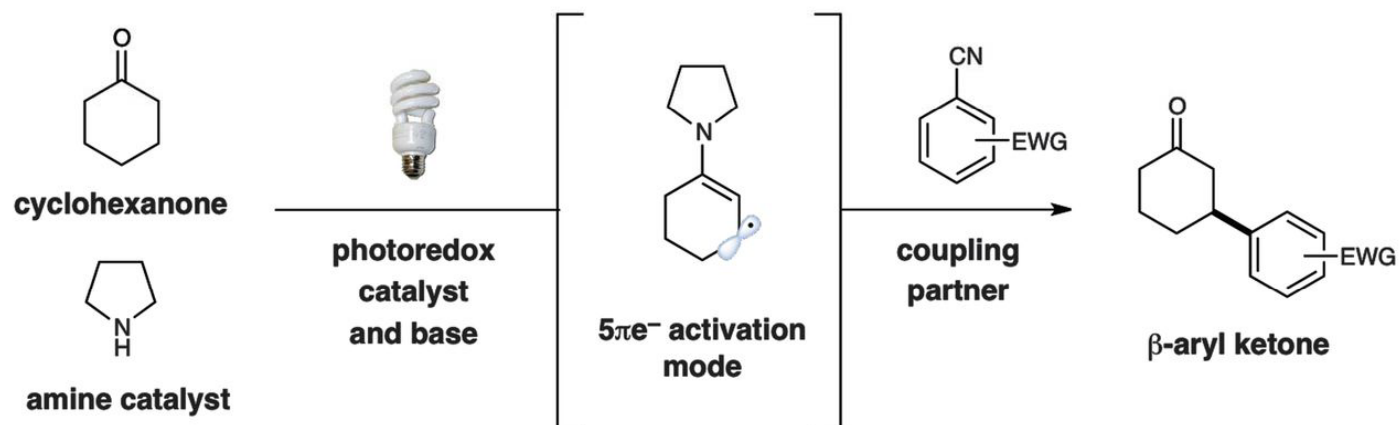
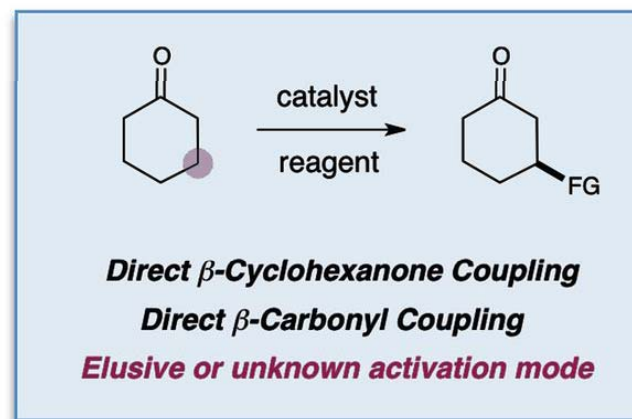
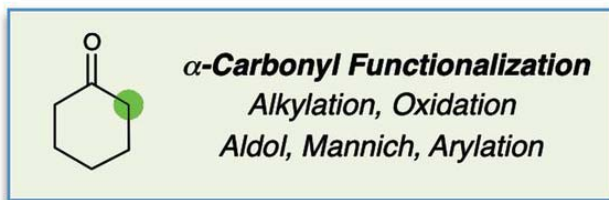
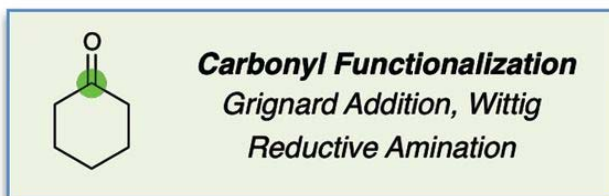
Shih, H.-W.; Vander Wal, M. N.; Grange, R. L.; MacMillan, D.W.C. *J. Am. Chem. Soc.* **2010**, *132*, 13600

Photoredox SOMO catalysis

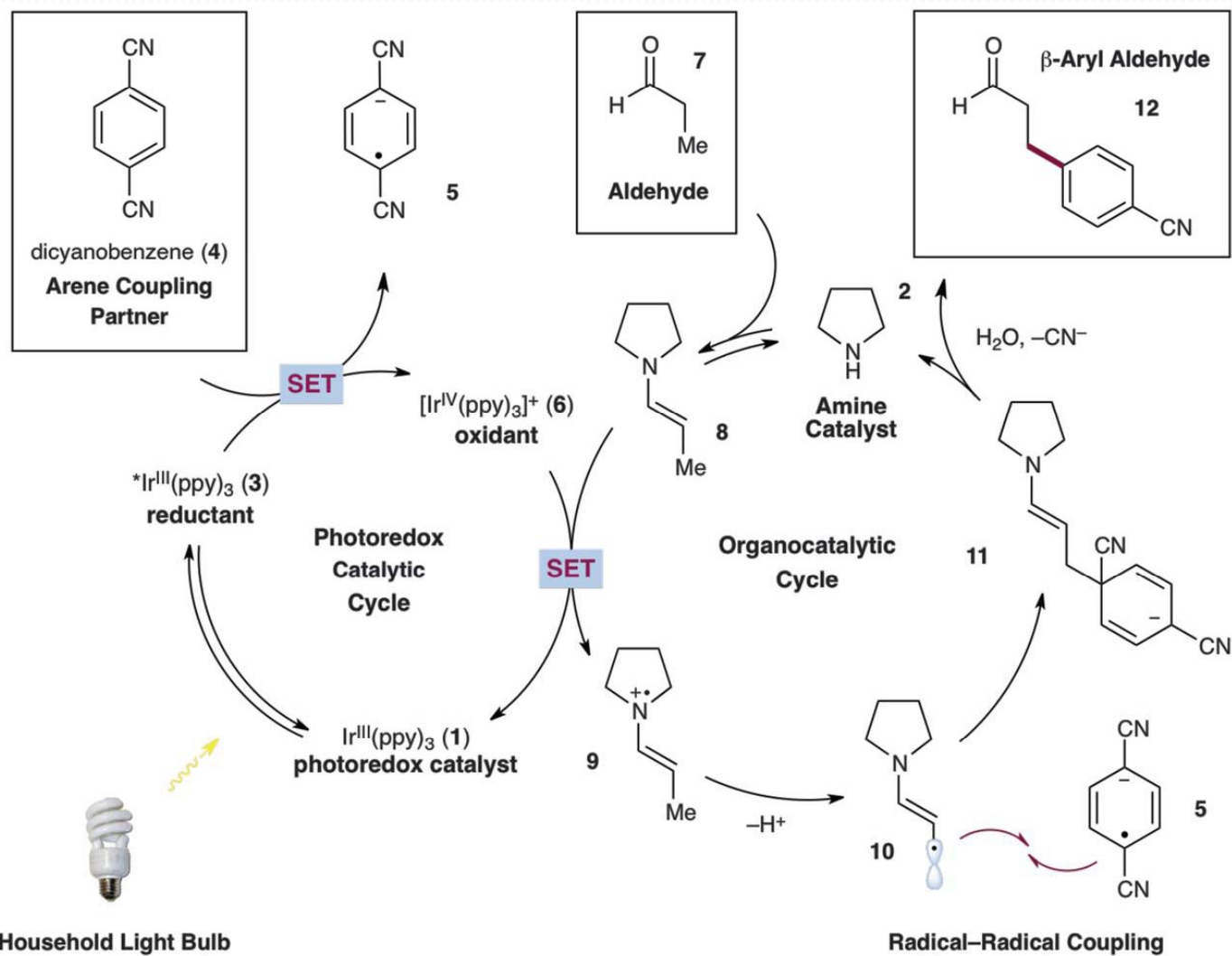
Enantioselective synthesis of a bioactive drug candidate



Photoredox SOMO catalysis

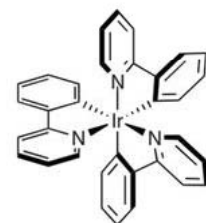
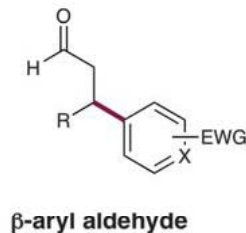
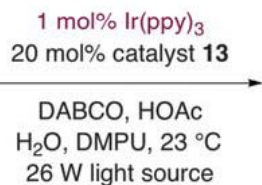
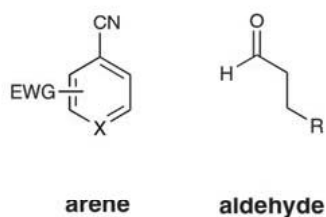


Photoredox SOMO catalysis



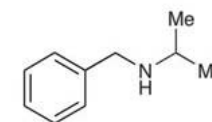
Pirnot, M. T.; Rankic, D. A.; Martin, D. B. C.; MacMillan, D.W.C. *Science*. **2013**, 339, 1593

Photoredox SOMO catalysis



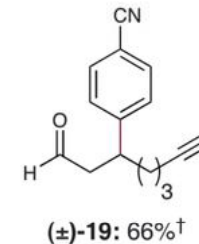
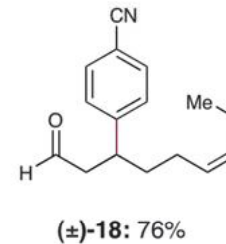
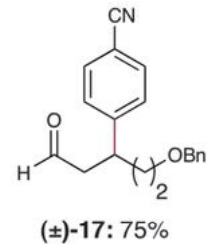
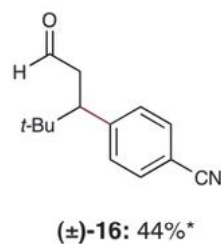
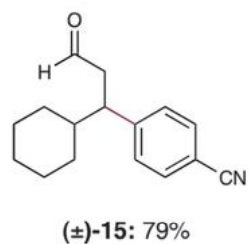
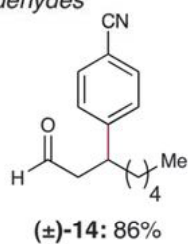
Ir(ppy)₃ (1 mol%)

Catalyst Combination

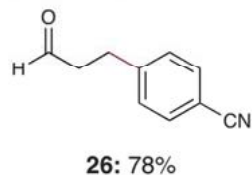


organocatalyst **13** (20 mol%)

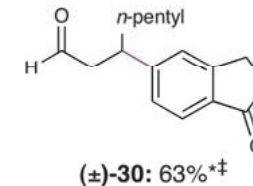
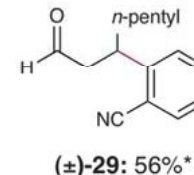
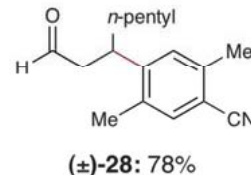
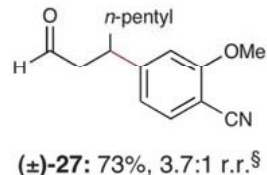
Aldehydes



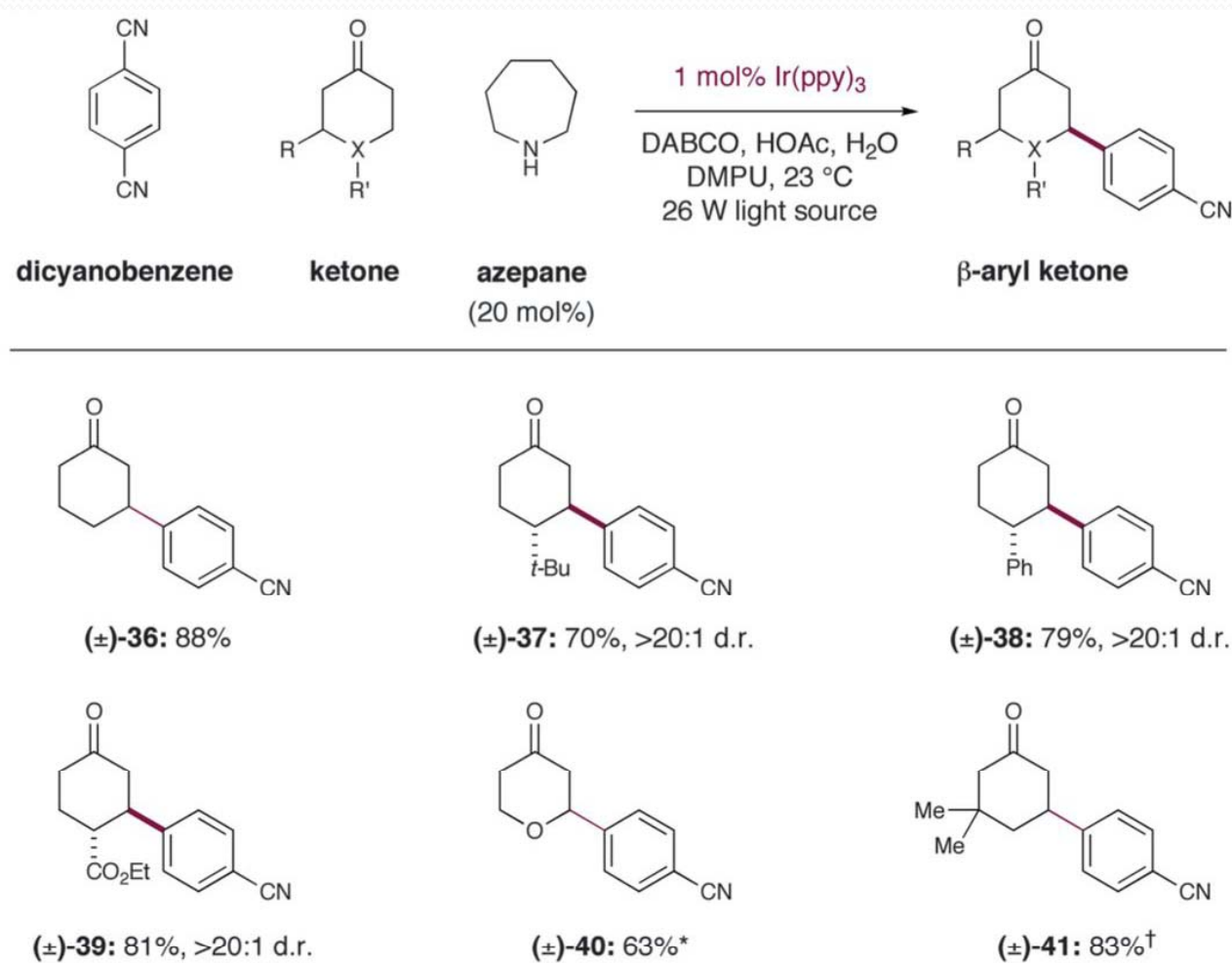
Propionaldehyde



Arenes

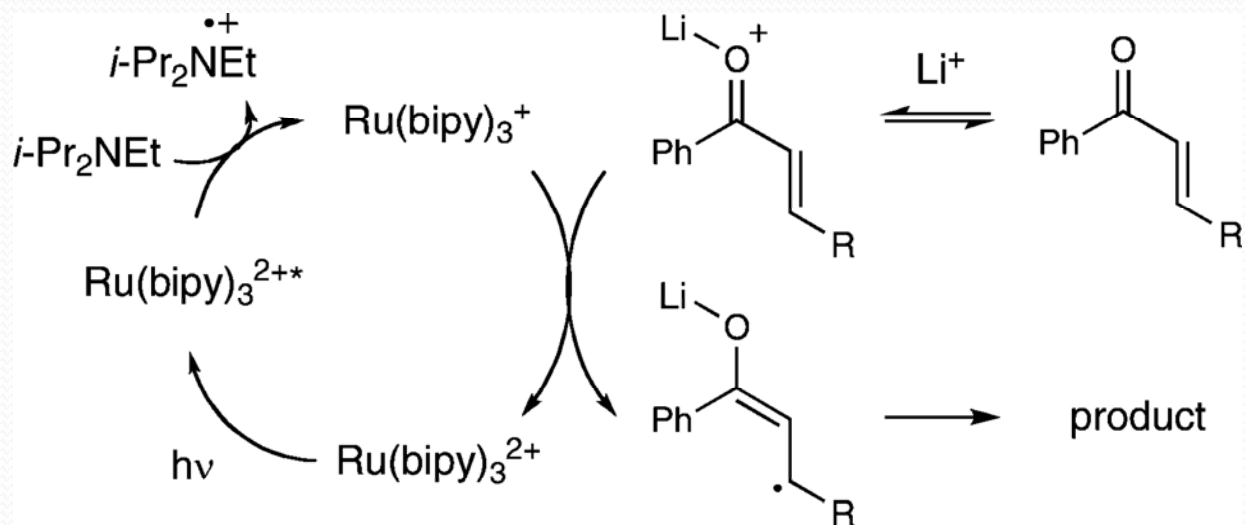
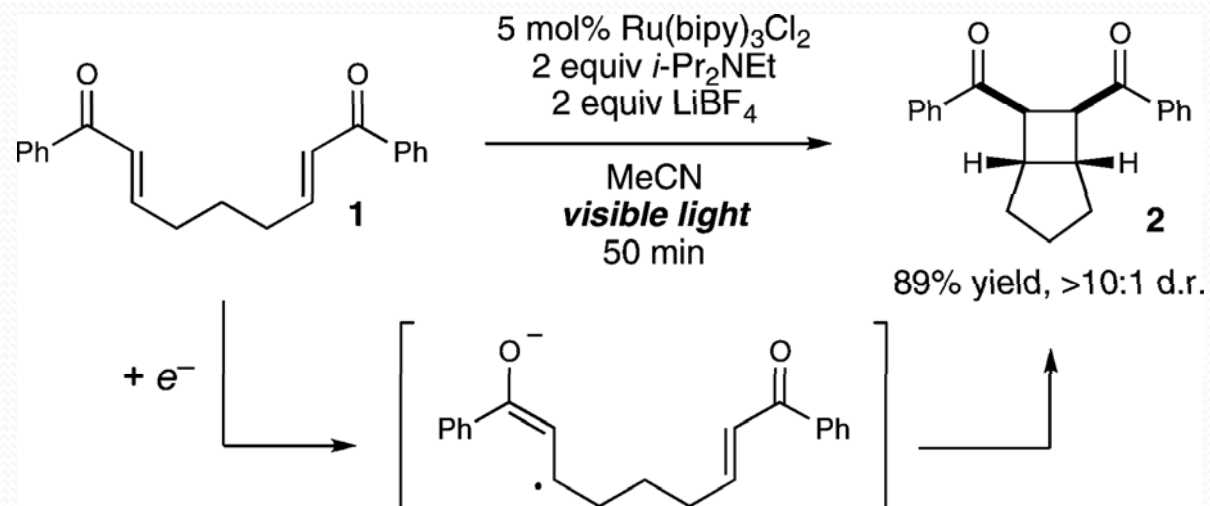


Photoredox SOMO catalysis



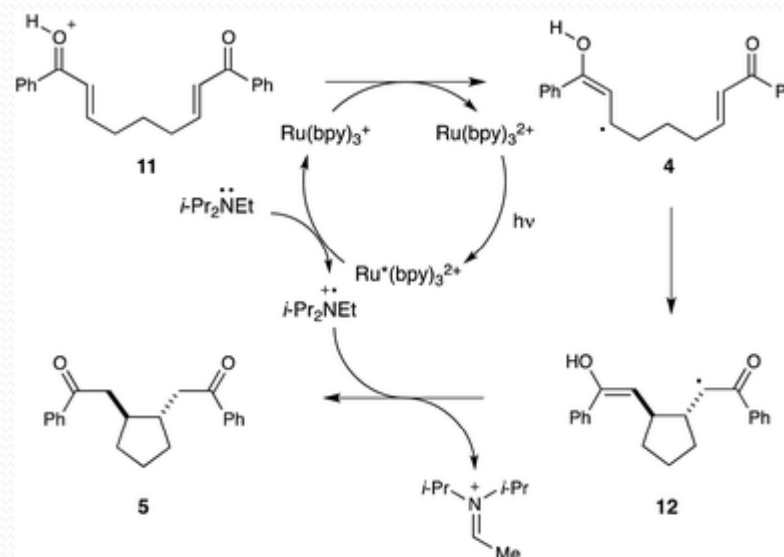
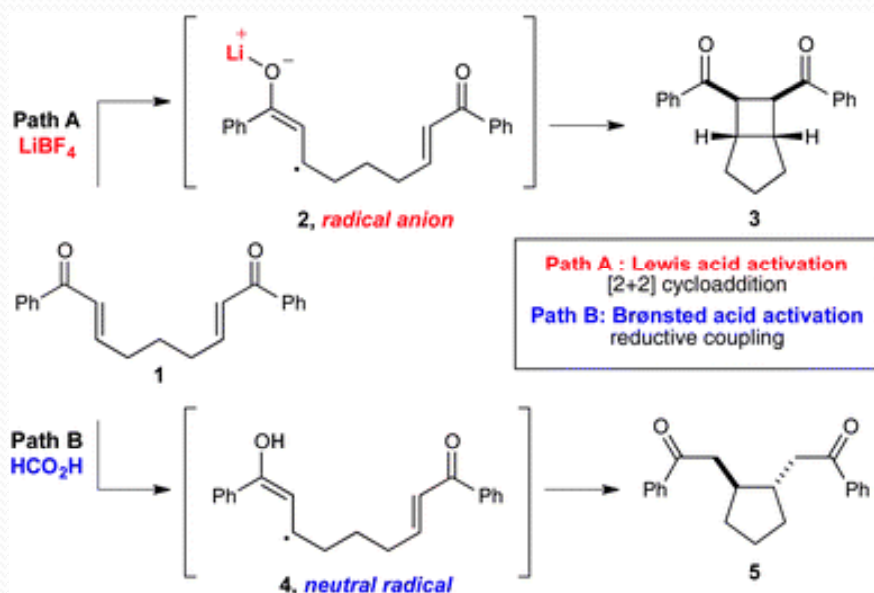
Pirnot, M. T.; Rankic, D. A.; Martin, D. B. C.; MacMillan, D.W.C. *Science*. **2013**, 339, 1593

Photoredox catalysis

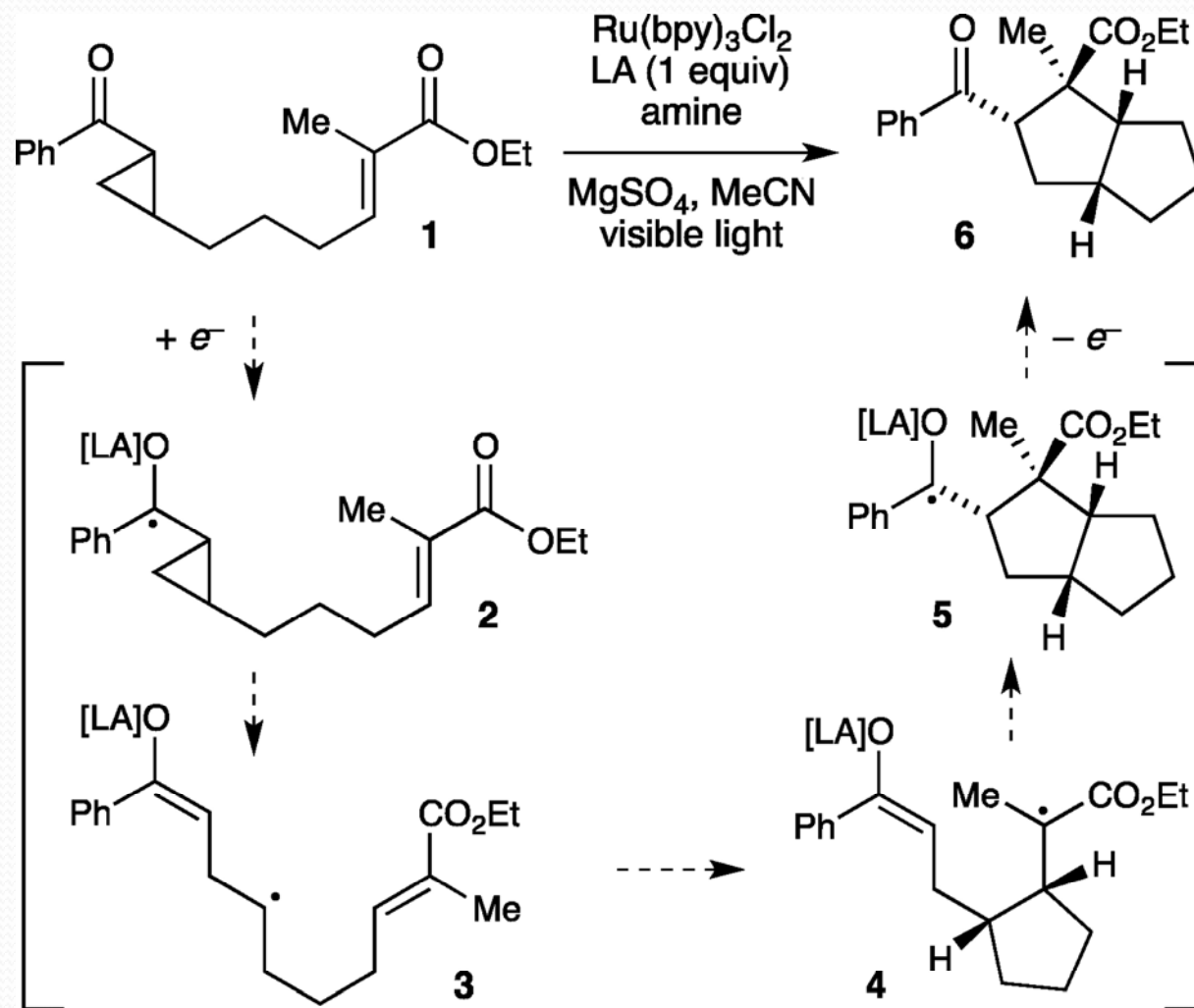


Ischay, M. A.; Anzovino, M. E.; Du, J.; Yoon, T. P. *J. Am. Chem. Soc.* **2008**, *130*, 12886

Photoredox catalysis

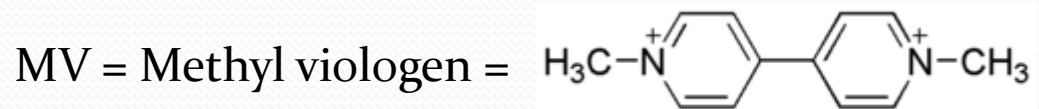
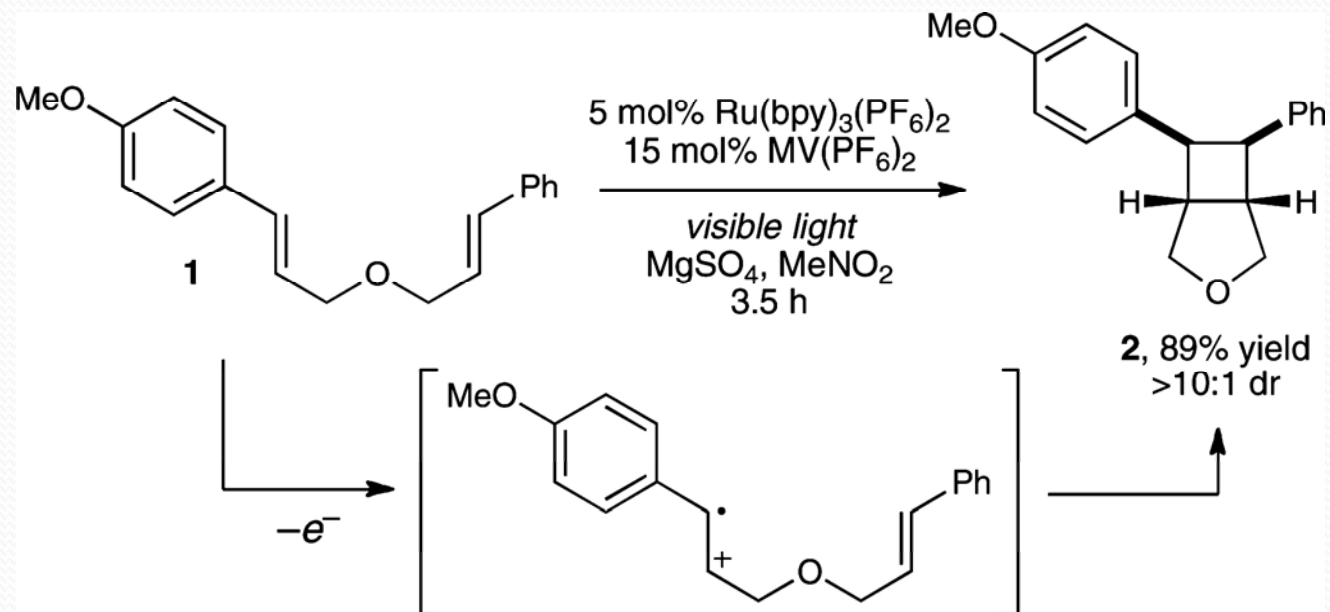


Photoredox catalysis



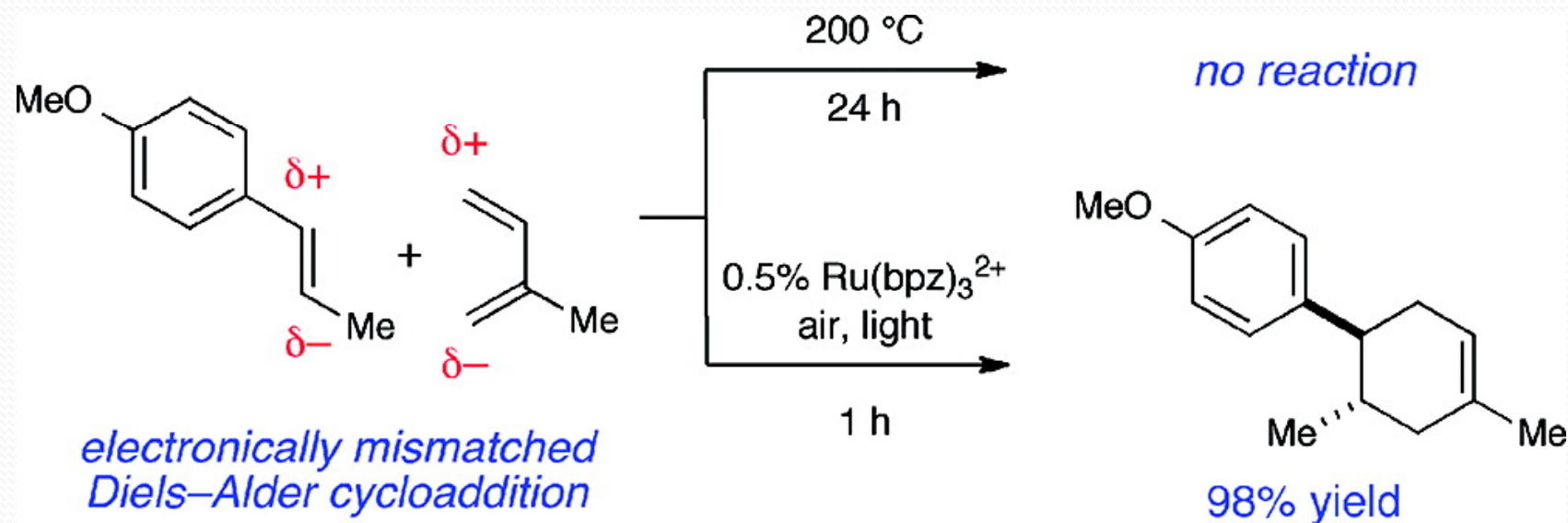
Lu, Z.; Shen, M.; Yoon, T. P. *J. Am. Chem. Soc.* **2011**, *133*, 1162

Photoredox catalysis

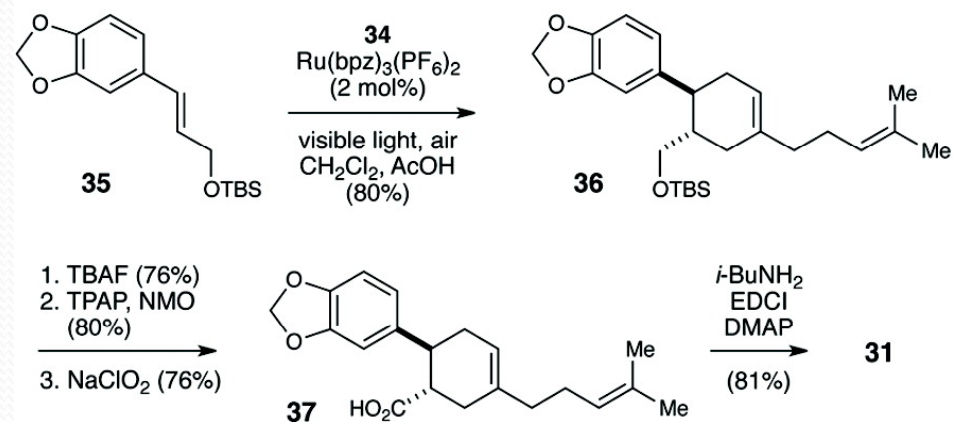
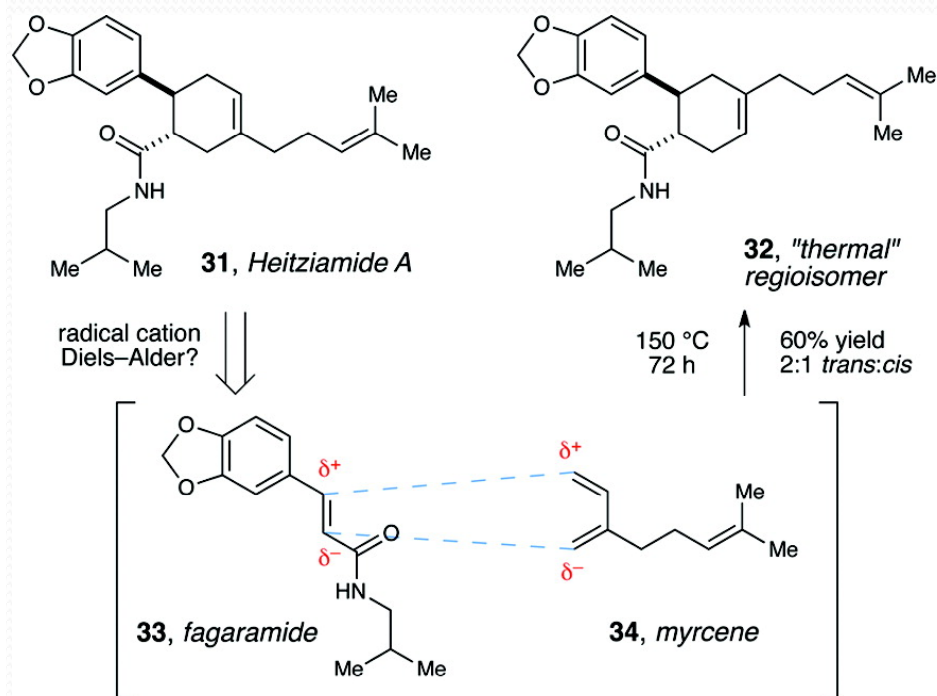


Ischay, M. A.; Lu, Z.; Yoon, T. P. *J. Am. Chem. Soc.* **2010**, *132*, 8572

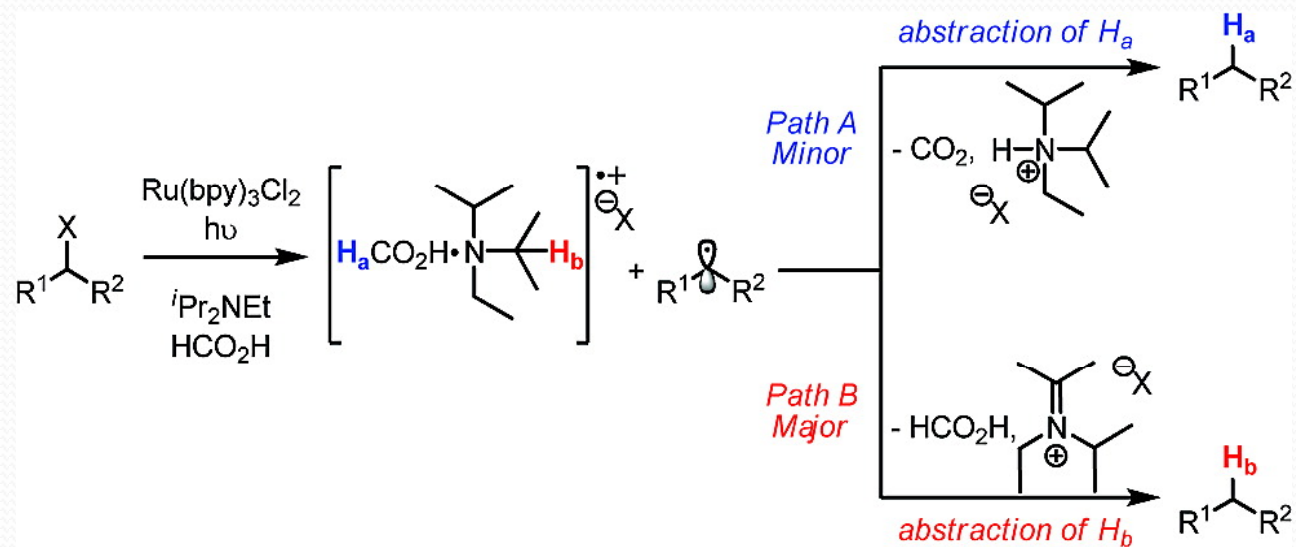
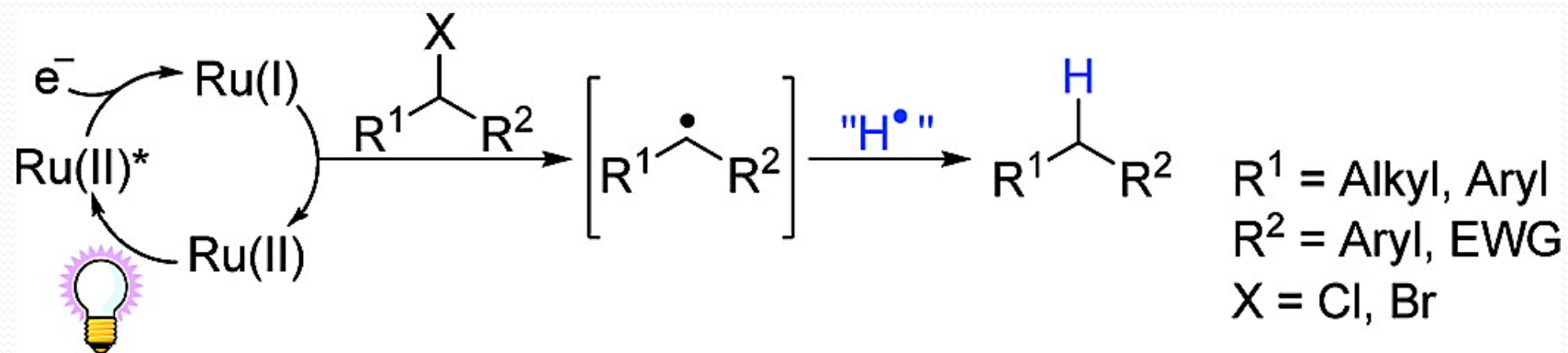
Photoredox catalysis



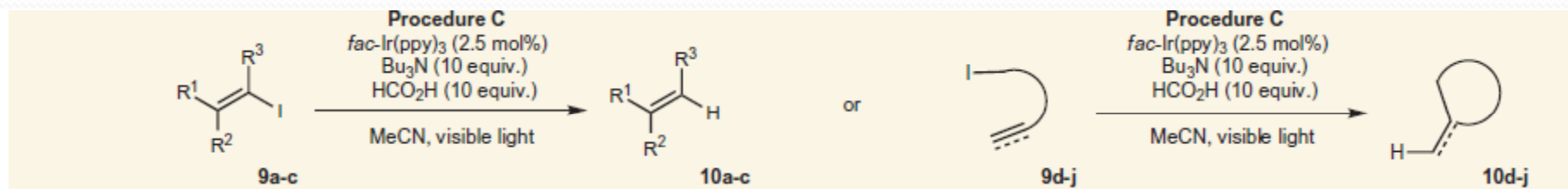
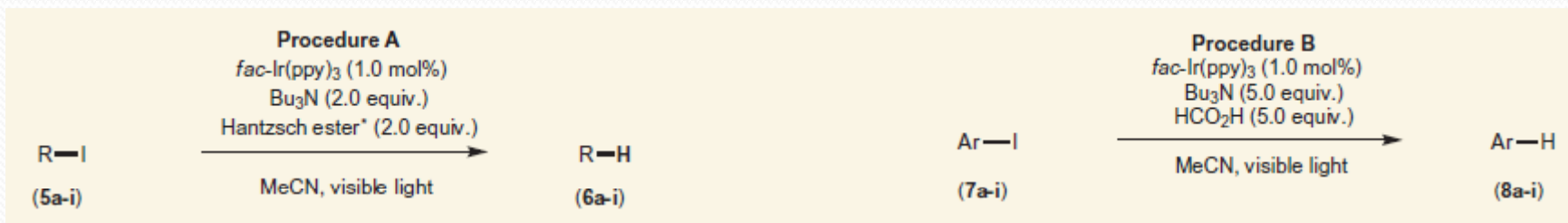
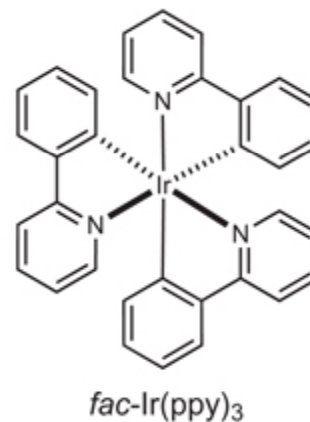
Photoredox catalysis



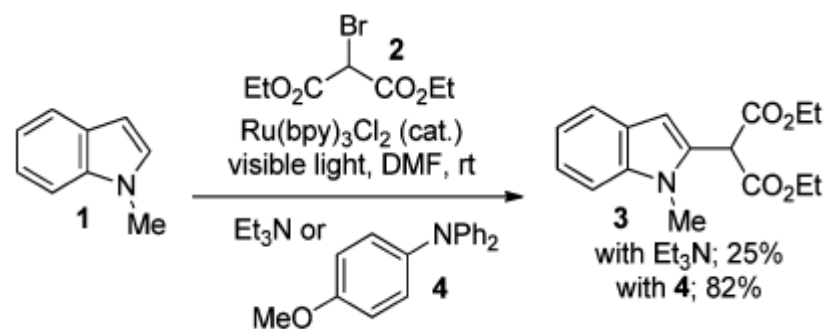
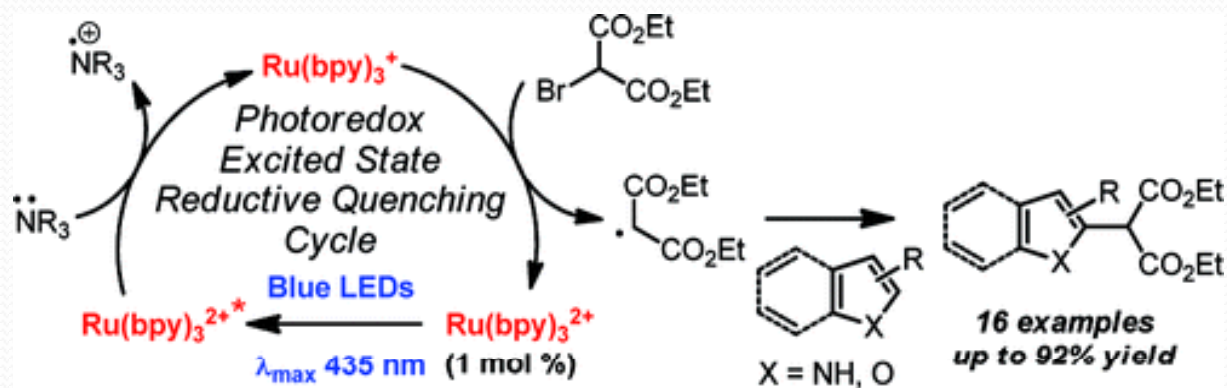
Photoredox catalysis



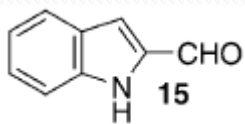
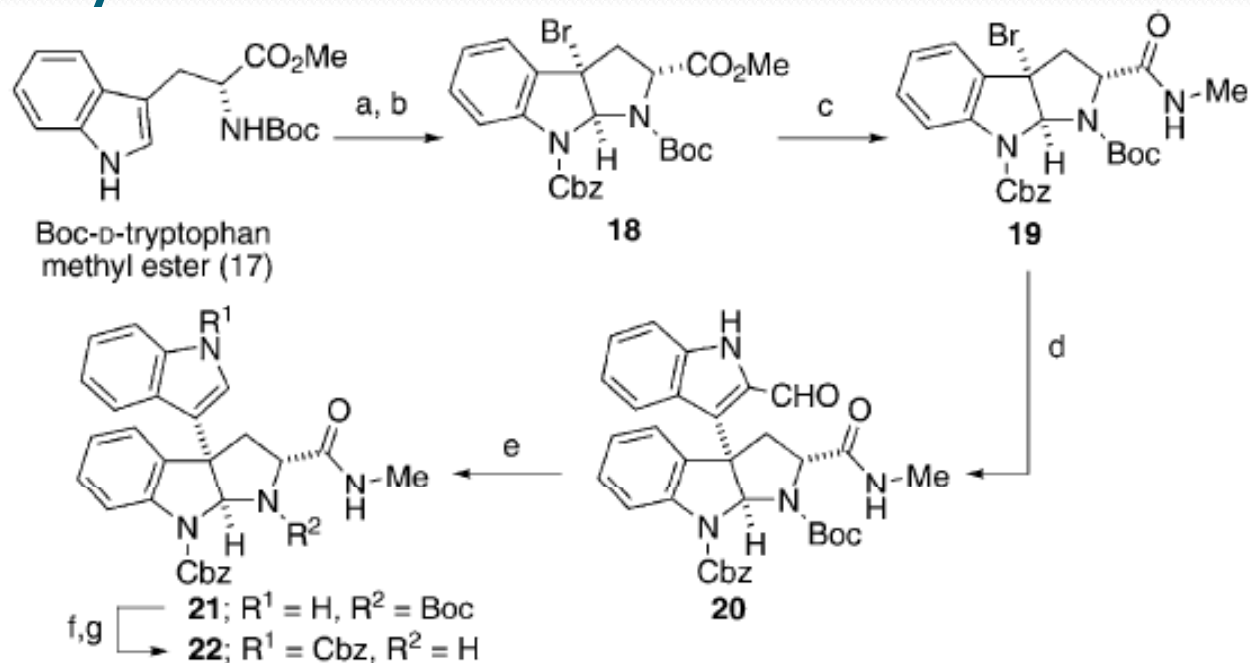
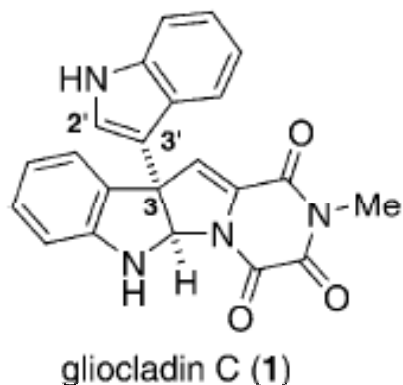
Photoredox catalysis



Photoredox catalysis

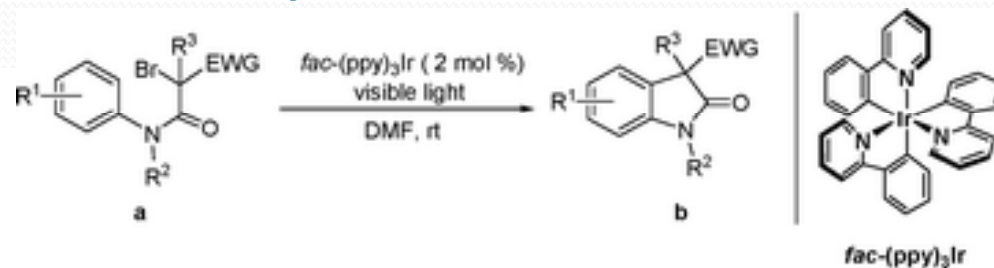


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Scheme 3. a) CbzCl, NaOH, Bu₄NHSO₄, CH₂Cl₂, 12 h; b) NBS, PPTS, CH₂Cl₂, 23 °C, 12 h, 91 % (two steps); c) MeNH₂, THF, 23 °C, 3 d, 87%; d) [Ru(bpy)₃Cl₂] (1.0 mol %), Bu₃N (2 equiv), 15 (5 equiv), DMF, blue LEDs, 12 h, 82%; e) [Rh(Ph₃P)₃Cl] (1 equiv), xylenes, 140 °C, 12 h, 86% or [Rh(CO)(Ph₃P)₂Cl] (20 mol %), dppp (44 mol %), DPPA (2 equiv), xylenes, 140 °C, 85%; f) CbzCl, NaOH, Bu₄NHSO₄, CH₂Cl₂, 12 h, 98%; g) TMSI, CH₃CN, 0 °C, 1 h, 91 %. Cbz = benzyloxycarbonyl;

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Entry	Catalyst	Additive (equiv.)	Conversion (%)	Yield (%) ^b
1	Ru(bpy) ₃ Cl ₂	Et ₃ N (2)	70	35
2	Ru(bpy) ₃ Cl ₂	ⁱ Pr ₂ NEt (2)	100	60
3	Ru(bpy) ₃ Cl ₂	Ph ₃ N (2)	17	14
4	Ru(bpy) ₃ Cl ₂	none	<1	N.D. ^c
5	Ir(ppy) ₂ (dtbpy)PF ₆	Ph ₃ N (2)	80	71
6	<i>fac</i> -(ppy) ₃ Ir	2,6-lutidine (2)	100	95
7	<i>fac</i>-(ppy)₃Ir	none	100	95
8	<i>fac</i> -(ppy) ₃ Ir	none	<1 ^d	N.D. ^c
9	none	none	<1	N.D. ^c

^a A solution of 0.12 mmol of **1a** and 2.4×10^{-3} mmol of catalyst in DMF (2.4 mL) was irradiated at room temperature under argon atmosphere for 12 h. ^b Isolated yield. ^c Not determined. ^d Control experiment without irradiation under the otherwise same conditions.

Conclusion

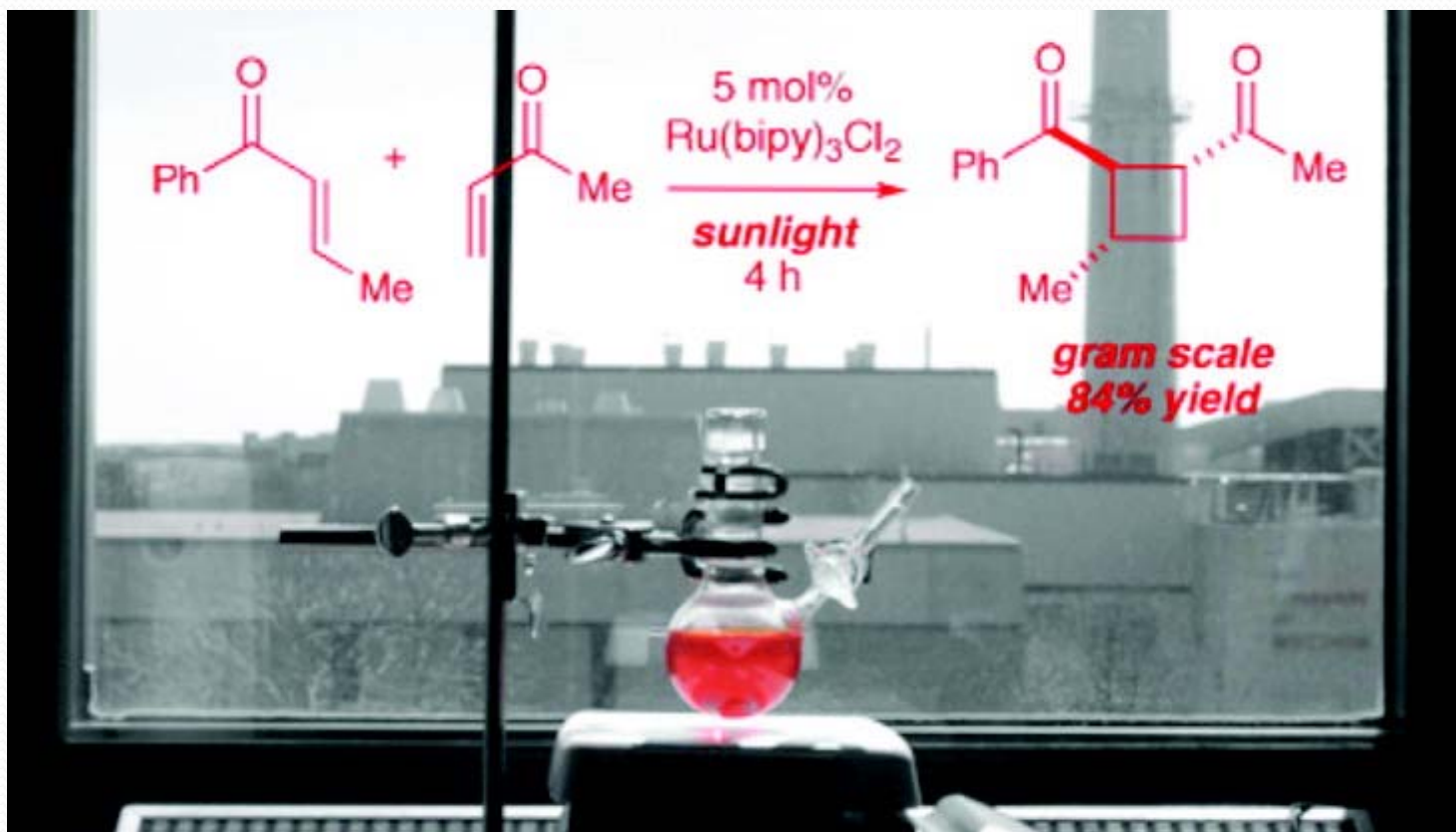
- Using SOMO catalysis many different building blocks for pharmaceutical and medicinal chemistry could be synthesised
- Photocatalyst such as Ru complexes can be used as one electron oxidizer
- Radicals are the most present intermediates in photocatalysis



Thank you for your attention

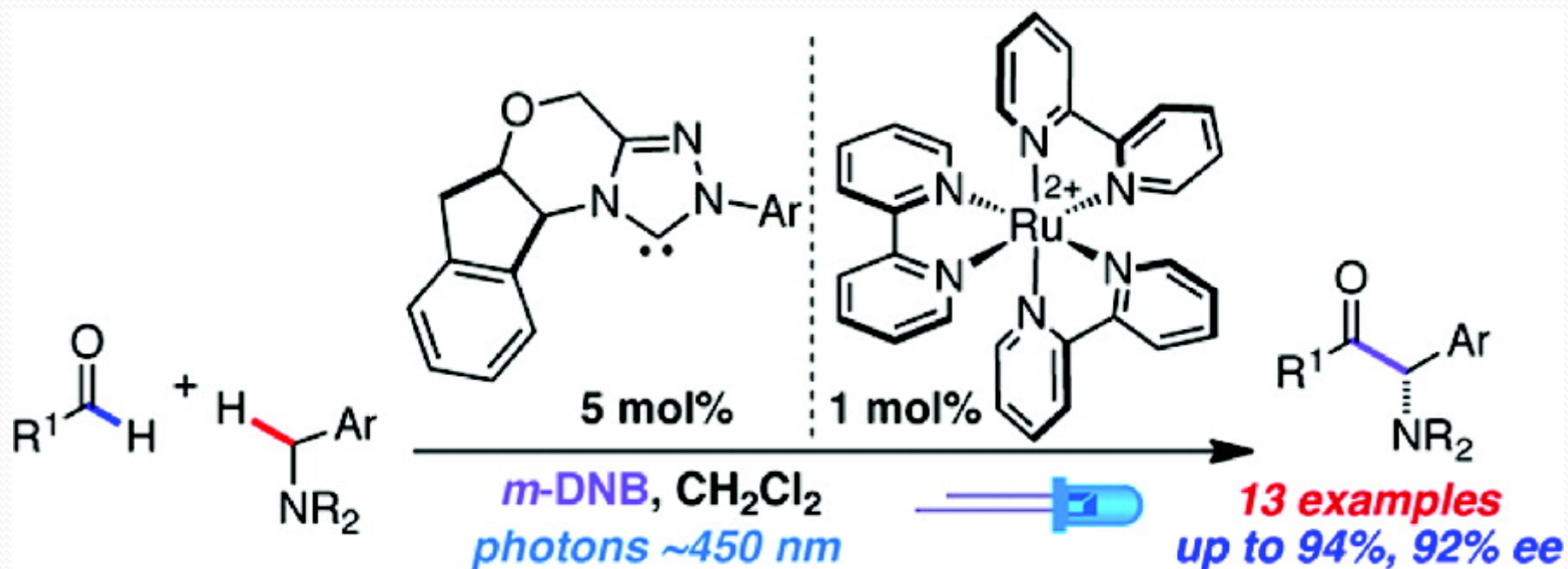


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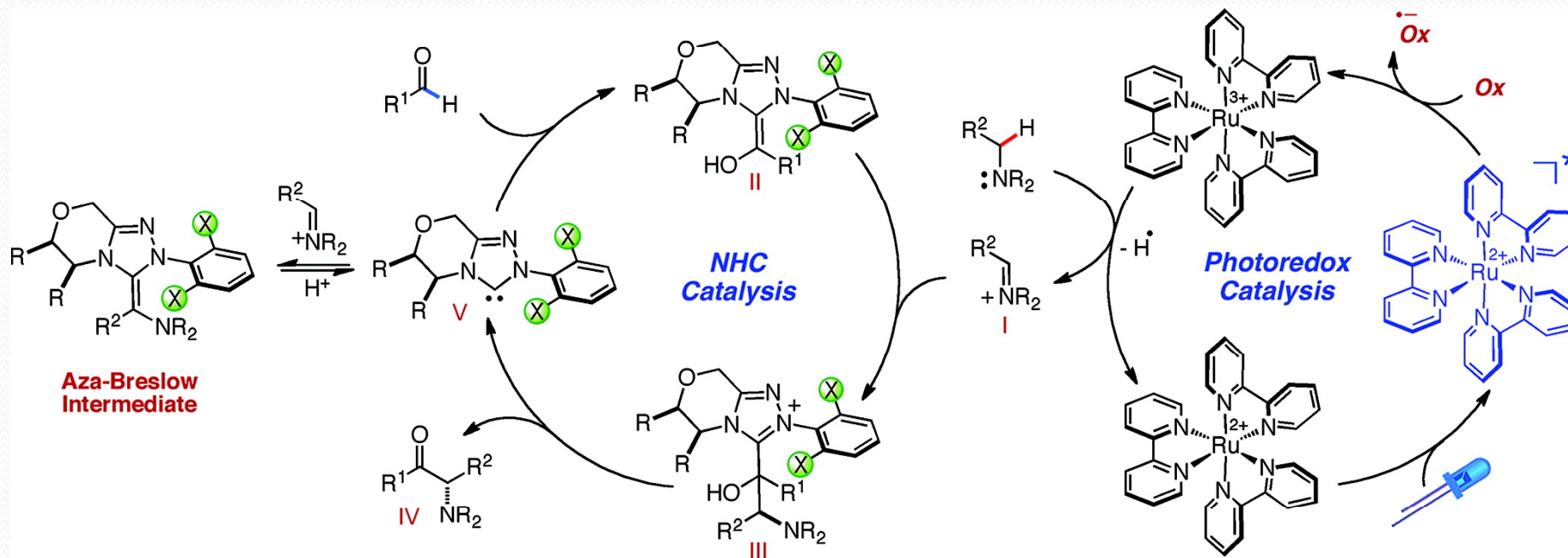


Du, J.; Yoon, T. P. *J. Am. Chem. Soc.* **2009**, *131*, 14604

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DiRocco, D. A.; Rovis, T. *J. Am. Chem. Soc.* **2012**, *134*, 8094